N-helix and Cysteines Inter-regulate Human Mitochondrial VDAC-2 Function and Biochemistry

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Human voltage-dependent anion channel-2 (hVDAC-2) functions primarily as the crucial anti-apoptotic protein in the outer mitochondrial membrane, and additionally as a gated bidirectional metabolite transporter. The N-terminal helix (NTH), involved in voltage sensing, bears an additional 11-residue extension (NTE) only in hVDAC-2. In this study, we assign a unique role for the NTE as influencing the chaperone-independent refolding kinetics and overall thermodynamic stability of hVDAC-2. Our electrophysiology data shows that the N-helix is crucial for channel activity, whereas NTE sensitizes this isoform to voltage gating. Additionally, hVDAC-2 possesses the highest cysteine content, possibly for regulating reactive oxygen species content. We identify interdependent contributions of the N-helix and cysteines to channel function, and the measured stability in micellar environments with differing physicochemical properties. The evolutionary demand for the NTE in the presence of cysteines clearly emerges from our biochemical and functional studies, providing insight into factors that functionally demarcate hVDAC-2 from the other VDACs.

The voltage-dependent anion channels (VDACs) are the most abundant outer mitochondrial membrane β-barrel proteins in eukaryotes. In humans, they exist as three isoforms (hVDAC-1, -2, and -3) and together govern the passage of metabolites and ions across the outer mitochondrial membrane (1, 2). hVDAC-1 is 10- and 100-fold more abundant than hVDAC-2, the NTH carries an additional 11-residue sequence that we term “N-terminal extension” (NTE, residues 1–11) (Fig. 1). NTE is absent in the other VDAC isoforms, and hVDAC-1 is 10- and 100-fold more abundant than hVDAC-2 and -3, respectively, and is the most characterized VDAC protein (1, 3). Crystal and NMR structures for hVDAC-1 are concurrent with a 19-stranded β-barrel, bearing an N-terminal helix (NTH) lining the pore interior (4–6). hVDAC-2 and hVDAC-3 share 72 and 69% sequence identity to hVDAC-1, and are expected to adopt similar structural scaffolds (1).

hVDAC-1 is known to exhibit the highest ion channel activity, which is voltage-gated by the NTH (5, 7). Furthermore, hVDAC-1 and -3 possess pro-apoptotic functions (1), whereas hVDAC-2 is gaining popularity for its anti-apoptotic property (inhibition of BAK function) (8). The recent crystal structure of zebrafish VDAC-2 revealed that the NTH lines the pore interior in a manner resembling hVDAC-1 (9). A definitive placement of this amphipathic helix in all VDAC isoforms has been the subject of constant debate. There is experimental evidence for the NTH lining the pore (4–6, 9–11), located outside the pore and toward the cytosol (12, 13) forming a part of the barrel wall (14, 15) and the remaining membrane adsorbed (16). It was seen that the NTH required a specialized hydrophobic or amphipathic environment to attain its structure in aqueous conditions (17–19). Taken together, the data reflect the highly dynamic nature of the NTH.

The hVDAC-1 NTH is essential during apoptosis and cytochrome c release, as it interacts with the anti-apoptotic factors Bcl-2 and hexokinase-1 (20). Conversely, NTH is dispensable for metabolite exchange, cell growth, mitochondrial targeting of the protein, as well as barrel oligomerization (17, 20). NTH deletion results in complete loss of VDAC voltage dependence (7, 20–24), suggesting that the NTH is a voltage-gating sensor. It is, however, unclear whether the NTH-less channel remains in the high conductance (20, 23) or the subconductance state (7, 21, 22, 24). Models that explain the NTH-mediated gating and hVDAC-1 subconductance state at ≥±30 mV include: (i) helix translocation into the barrel, causing pore closure (25); (ii) alteration of NTH-pore interaction due to change in NTH structure (26, 27); (iii) translocation of the helix from the pore wall to the center (5, 6); (iv) ellipsoidal barrel formation upon NTH removal (7); and (v) no structural rearrangement during gating (28).

It is evident that the NTH of VDAC-1 can possess diverse functional properties and structural characteristics. In hVDAC-2, the NTH carries an additional 11-residue sequence that we term “N-terminal extension” (NTE, residues 1–11) (Fig. 1A). NTE is absent in the other VDAC isoforms, and is implicated in hVDAC-2 channel activity, as well as resistance to reactive oxygen species (ROS) (29). The role of the NTH and...
NTE of hVDAC-2 were recently examined in the context of channel function and gating (24, 29, 30). However, our understanding of the N-helix (NTH + NTE) function in hVDAC-2 is remarkably poor when compared with hVDAC-1. Furthermore, the possibility that the helix plays additional structural roles in hVDAC-2 has not yet been examined.

In this study, we address the role of hVDAC-2 NTH and NTE in the function and stability of this transmembrane β-barrel. Furthermore, we evaluate the significance of our findings in micellar systems and lipid bicelles. We demonstrate that the removal of NTE and NTH disrupts barrel voltage sensing and stability in a manner that is considerably influenced by the presence of the hVDAC-2 cysteine residues. Additionally, our data supports a biophysical role for the N-helix in barrel refolding kinetics and points to an evolutionary significance for why hVDAC-2 requires an extended N-terminal segment.

**Experimental Procedures**

**Production and Refolding of hVDAC-2 WT (Wild Type) and Its Mutants—**Human vdac-2 gene in pET-3b (31) was used as the template for the generation of c0-pET-3b, through site-directed mutagenesis. hVDAC-2 constructs lacking NTE (Δ1–11) and NTE + NTH (Δ1–32) were cloned from these parent constructs. Cys-less full-length (FL C0) had the following mutations: C8S, C13A, C47E, C76T, C103A, C133H, C138S, C210N, and C227N (see Fig. 1C). *Escherichia coli* BL21(DE3) cells were transformed with these plasmids, and the proteins were synthesized by Fmoc-ω-(9-fluorenyl)methoxycarbonyl chemistry, using established protocols (34), with a Cys-13 → Ala replacement. Two sample types with different protein:peptide ratios (1:10–1:1000) were prepared: (i) peptide was added to the unfolded protein and co-refolding was done; (ii) peptide was separately folded in LDAO by mixing, and added to the refolded protein stock, or directly in the cis and trans chamber, whereas carrying out the planar lipid bilayer experiments.

**Barrel Refolding Kinetics Measurements—**Refolding kinetics were monitored using Trp fluorescence anisotropy ($r$) by rapid 10-fold dilution of 250 μM denatured protein (in 6 M GdnHCl and 10 mM DTT) in 65 mM LDAO (lauryldimethylamine oxide) and 19.5 mM DDM (dodecyl β-D-maltoside) prepared in Buffer A (100 mM NaCl, 10 mM DTT, 50 mM phosphate buffer, pH 7.2) at 4 °C. Refolding reactions were incubated at 4 °C for 5 h, and aggregated protein, if any, was removed by centrifugation at 15,500 $g$ for 1 h at 4 °C. The final refolded protein 5× stocks thus prepared contained 25 μM protein in 65 mM LDAO or 19.5 mM DDM in Buffer A; residual GdnHCl (0.6 M) was not removed.

Bicelles of varying $q$ ($q = \frac{\text{[long chain lipid; DMPC]}}{[\text{short chain lipid; LDAO or DDM}]}$) ranging from 0.00015 to 1, were generated by subjecting a DMPC (1,2-dimyristoyl-sn-glycero-3-phosphocholine) gradient (0.01 to 20 mM), prepared in 65 mM LDAO or 19.5 mM DDM in Buffer A, to five freeze-thaw cycles (31). 250 μM denatured protein (in 6 M GdnHCl and 10 mM DTT) was refolded by 10-fold dilution in the bicelles preparations, as described for the micelles. Samples were additionally dialyzed against Buffer A to remove residual GdnHCl and centrifuged to remove aggregated protein, to generate the 5× refolded stock.

Unless specified otherwise, all stock proteins were diluted 5-fold in Buffer A (without DTT), for the various measurements.

**Channel Conductance Measurements—**Refolded protein (5× refolded protein stock (25 μM) in 65 mM LDAO in Buffer A supplemented with 0.1% cholesterol and 1% Triton X-100 (32)) was added to the cis side of a planar lipid bilayer made up of 12.5 mg/ml of DiPhPC (diphytanoyl phosphatidylcholine) + 0.1% cholesterol painted across a 200-μm aperture. The cis and trans chambers contained 10 mM HEPES buffer, pH 7.4, 5 mM CaCl$_2$, and 1 M KCl (32). All bilayer recordings and data analyses were carried out using reported methods (33). Briefly, the channel insertion events were recorded at a holding voltage of $+10$ mV with a filtering frequency of 400 Hz and voltage dependence was studied using a 3 mV triangular voltage ramp between ±60 mV at a filtering frequency of 15 kHz. 20–100 active channels were used for voltage ramp measurements.

The conductance ($G/G_{\text{max}}$) and open probability ($P_{\text{open}}$) plots were calculated for the channel reopening part of the ramp (33). The steepness of the voltage dependence ($n$) and the voltage at which half the channels are open ($V_0$) were obtained using Equation 1 (33).

$$\ln \left[ \frac{G_{\text{min}} - G}{G - G_{\text{max}}} \right] = -\frac{nFV + nFV_0}{RT}$$

(Eq. 1)

Here, $G$ is the conductance at any voltage $V$, $G_{\text{min}}$ and $G_{\text{max}}$ are the minimum and maximum conductance seen at higher and lower voltages, respectively, $F$ is the Faraday constant, $R$ is the gas constant in joules, and $T$ is the temperature in kelvin.

**Peptide Synthesis and Sample Preparation—**The NTH sequence of hVDAC-2 WT (residues 12–32; see Fig. 1B), was synthesized by Fmoc-ω-(9-fluorenyl)methoxy carbonyl chemistry, using established protocols (34), with a Cys-13 → Ala replacement. Two sample types with different protein:peptide ratios (1:10–1:1000) were prepared: (i) peptide was added to the unfolded protein and co-refolding was done; (ii) peptide was separately folded in LDAO by mixing, and added to the refolded protein stock, or directly in the cis and trans chamber, whereas carrying out the planar lipid bilayer experiments.

**Equilibrium Unfolding and Refolding—**Intrinsic Trp fluorescence was used to measure equilibrium unfolding in GdnHCl (35). 25 μM protein in 65 mM LDAO or 19.5 mM DDM in Buffer A (refolded stock) or Buffer A supplemented with 6 M GdnHCl (for LDAO) or 4 M GdnHCl (for DDM) (unfolded stock) were diluted 5-fold in a GdnHCl gradient. The final reactions contained 5 μM protein, 13 mM LDAO, or 3.9 mM DDM in Buffer A containing 2 mM DTT. The progress of all the reactions were monitored for 16 h in LDAO and 72 h in DDM using $\lambda_{\text{em}} = 295$ nm and $\lambda_{\text{em}} = 340$ nm were used to acquire data every ~15.74 s. The experiment dead time was ~20–25 s (31). Data were fitted to a single exponential function to derive the refolding rate ($k_r$) (31).
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\((\Delta G_{\text{app}}^{u})\) and apparent unfolding cooperativity \((m_{\text{app}}^{u})\) were derived (36).

Anisotropy, Quenching, and Lifetime Measurements—Trp fluorescence anisotropy \((r)\) was acquired at \(\lambda_{\text{ex}} = 295 \text{ nm}\) using \(\lambda_{\text{em}} = 340 \text{ nm}\), slit width of 5 nm, and integration time of 5 s over three trials to achieve a standard error <2% (31). Acrylamide quenching measurements of Trp fluorescence were recorded at 25 °C using established protocols (31), to derive the Flory-Huggins parameter \((K_{\text{FH}})\). Trp lifetimes were measured using time-correlated single photon counting using \(\lambda_{\text{ex}} = 292 \text{ nm}\) and the data were collected at 340 nm. Fits of the data to a triple exponential function provided amplitude fractions \((\alpha_i)\) and corresponding lifetimes \((\tau_i)\). The average lifetime \((<\tau>)\) was derived using Equation 2.

\[
<\tau> = \alpha_i \tau_i \quad \text{(Eq. 2)}
\]

Bimolecular quenching constant \((q)\) and apparent rotational correlation time \((\tau_r)\) were calculated using Equations 3 and 4.

\[
k_q = K_{\text{SV}}/<\tau> \quad \text{(Eq. 3)}
\]

\[
\tau_r = <\tau> r/(r_0 - r) \quad \text{(Eq. 4)}
\]

Here \(r_0\) was taken as 0.3 for indole moiety.

Thermal Denaturation and Protein Stability—Far-UV circular dichroism (CD) wavelength scans were obtained using a scan rate of 100 nm/min, 0.1 cm path length, 1 nm bandwidth, and 1 s response time, at 4 °C. Raw CD values \((\theta)\) in degrees were converted to mean residue ellipticity \((\text{MRE}; \theta_{\text{MRE}})\) using Equation 5.

\[
\theta_{\text{MRE}} = \frac{\text{MRW} \times \theta}{10 \times d \times c} \quad \text{(Eq. 5)}
\]

Here, MRW = \(M(N-1)\), \(M\) is molecular weight of the protein in Da and \(N\) is the number of residues in the protein, \(d\) is the path length in cm, and \(c\) is the protein concentration in g/ml.

Thermal denaturation was monitored at 215 nm (MRE215) at a ramp rate of 1 °C/min, with sampling carried out at every 1 °C from 4 to 95 °C (31). Data were converted to \(f_U\) (31) and fitted with a two-state equation (37) to derive \(T_m\) (midpoint of thermal denaturation) and unfolding cooperativity. For thermal denaturation in DMPC:LDAO and DMPC:DDM systems, the refolded protein stocks were diluted 5-fold in Buffer A (without DTT) to achieve a DMPC gradient of 2–4000 \(\mu\text{m}\), in 13 mM LDAO or 3.9 mM DDM containing 5 \(\mu\text{m}\) protein. These samples were subjected to thermal denaturation, and \(f_U\) was calculated (31). The \(T_m\) (temperature at which thermal denaturation is initiated) was derived using reported methods (35).

Differential Scanning Calorimetry Measurements—Phase transition temperatures of DMPC in vesicular and bicellar systems were monitored using Microlc VP-differential scanning calorimetry microcalorimeter. DMPC bicelles (2–4000 \(\mu\text{m}\)) in 13 mM LDAO or 3.9 mM DDM were made without DTT as described in the previous section. The samples were subjected to thermal denaturation from 4 to 60 °C, at a scan rate of 60 °C/h and acquisition was carried out in high gain mode using 1 s filtering. The data were blank corrected, normalized, and processed using Microlc Origin software.

Data Analysis and Validation—All activity measurements were averaged over at least 4 independent recordings and spectroscopic measurements at least twice, using freshly refolded samples. Unless otherwise specified, graphs represent average data, and the error bars have been omitted for clarity. Parameters \((T_m, T_m^{\text{start}}, m, n, V_{\text{app}})\) have been derived from fits to independent datasets, with the error bars representing the standard deviation between independent experiments. The thermodynamic parameters \((C_{1/m}, m, n, V_{\text{app}})\) were derived from fits to averaged data with the error bars representing the goodness of fit.

Results

hVDAC-2 \(\Delta^{1–32}\) Shows Loss of Voltage Dependence in Lipid Bilayers and \(\Delta^{1–11}\) Barrel Is Voltage Sensitized—The role of NTH in voltage sensing of hVDAC-1 is well established (7, 20, 22, 24). However, the requirement of the additional 11-residue NTE in hVDAC-2 is unclear. In the I-TASSER-predicted model of hVDAC-2 (Fig. 1A), NTE exists as a random coil, and this region may form minimal interactions with the channel wall. NTE may play a role in protection from ROS (29), but its role in regulating channel conductance, voltage gating, or acting as a docking site for the Bcl-2 family of proteins is unclear. We therefore probed the functional and physiological properties of hVDAC-2 WT Fl and the N-helix truncations in black lipid membranes, by designing hVDAC-2 constructs \(\Delta^{1–11}\) and \(\Delta^{1–32}\) (Fig. 1, A and B). Additionally, hVDAC-2 is known to have unusually high cysteine content. Cysteines are rare in proteins, and their unusual abundance in hVDAC-2 suggests that they may be functionally important. Hence, we assessed the interplay of cysteines and NTH(+NTE), by generating C0 constructs (Fig. 1C).

We used multichannel membranes to monitor the response of hVDAC-2 to triangular voltage ramps (33). In VDAC-1, separate gating mechanisms at positive and negative voltages gives rise to asymmetric voltage dependence in DiPhPC bilayers (38). In vivo function for this asymmetry is unknown. Bilayer properties, strand deletions, disulphides, and NTH mutations affect this asymmetry, through directed insertions, conformational changes, or gating charge neutralization (11, 23, 39, 40). In hVDAC-2 FL, a quantitative comparison of the normalized conductance (Fig. 2, A and B) reveals asymmetric gating, whereas the deletion of NTE \((\Delta^{1–11})\) elicits a more symmetrical response to the applied voltage (Fig. 2B). NTE deletion changes the voltage dependence in the positive arm for WT and negative arm for C0 (Fig. 2B). Hence, depending on the presence of cysteines, NTE deletion shows opposite effects on hVDAC-2, with an overall similar outcome of symmetrized voltage sensing. Based on the nFV\(_{\text{app}}\) values (discussed below), the removal of NTE perturbs hVDAC-2 asymmetric gating through a conformational change in the barrel.

Complete removal of NTE + NTH \((\Delta^{1–32})\) abolishes the voltage dependence of hVDAC-2 (Fig. 2A), similar to previous observations for other VDACs (7, 20–24). Supplementing a synthetic NTH separately to the constructs (Fig. 1A) reveals an overall similar outcome of symmetrized voltage sensing.
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Our single channel conductance measurements indicate that hVDAC-2 WT and C0 insert in DiPhPC membranes in mainly two different conductance states of \( \sim 2 \) nS (subconductance) and \( \sim 4 \) nS (open) (Fig. 3 and Table 1). The ratio of these states remains similar across FL and \( \Delta^{1-11} \) constructs. Surprisingly, N-helix deleted \( \Delta^{1-32} \) WT channel also shows the existence of

FIGURE 1. hVDAC-2 N-terminal deletion mutants and detergent systems used in this study. A, the I-TASSER (57) modeled structure of the hVDAC-2 barrel (gray) is shown on the left, with NTE in blue and NTH in red. Middle and right panels show the I-TASSER modeled structure of hVDAC-2 \( \Delta^{1-11} \) and \( \Delta^{1-32} \) WT, respectively. B, multiple sequence alignment of the N-terminal mutants of hVDAC-2, and color coded according to A. C, comparison of hVDAC-2 WT and C0 sequences, highlighting the nine cysteines of WT in yellow, and showing the corresponding replacements in C0. All residues have been color coded according to their physicochemical properties. D, schematic representation of the hVDAC-2 barrel in LDAO (upper panel) and DDM (lower panel) micelles. The hydrophobic core of the micelles is shown in light blue and the polar headgroups in dark blue. The hydrophobic core of the protein is demarcated by orange lines. Note the spherical and oblate shapes of LDAO and DDM micelles, respectively. Chemical structures of LDAO and DDM are also provided to highlight the size of the headgroups.

FIGURE 2. Channel conductance measurements of hVDAC-2 and its mutants in lipid bilayers. A, \( G/G_{\text{max}} \) plots for WT proteins (left panels) and C0 constructs (right panels), obtained in response to a voltage gradient ranging from \( \pm 60 \) to \( \pm 60 \) mV in DiPhPC + 0.1% cholesterol membrane. The data shows the importance of the NTH in voltage gating of hVDAC-2. In both the WT and C0 barrels loss of voltage dependence is seen upon deletion of the N-helix. WT constructs: FL (square), \( \Delta^{1-11} \) (triangle), and \( \Delta^{1-32} \) (hexagon); C0 constructs: FL (circle), \( \Delta^{1-11} \) (diamond), and \( \Delta^{1-32} \) (inverted triangle). B, comparison of the \( G/G_{\text{max}} \) plots of WT and C0 mutants of FL hVDAC-2 (left panel) and \( \Delta^{1-11} \) hVDAC-2 (right panel). C, the NTH peptide is unstructured (random coil conformation) in both water and Buffer A, and shows a helical conformation only in 65 mM LDAO (in Buffer A), when measured using far-UV CD. The data suggests that NTH can interact with the lipid environment in hVDAC-2, which can promote NTH structuring. D, exogenously supplemented NTH peptide is unable to restore voltage-dependent gating in \( \Delta^{1-32} \) mutants. Experiments were carried out with protein:peptide ratios ranging from 1:10 to 1:1000 and results for 1:1000 ratio is shown. Error bars in panels A, B, and D are derived from at least four independent experiments.
open and subconductance states, unlike hVDAC-1, where only the subconductance state is reported (7, 21, 22, 24). The Cys-rich Δ1–32 hVDAC-2 preferably exists in this subconductance state (Fig. 3 C and Table 1). Our results show that the open channel state is accessible to helix-less hVDAC-2 when cysteines are present. Loss of the N-terminal helix in Δ1–32 WT may not always give rise to the ellipsoidal barrel with a narrow pore suggested for hVDAC-1 (7), and may be accessible predominantly to Δ1–32 C0.

We calculated \( nFV_\sigma \), which indicates the energy difference between the open and closed channel states (33). This value decreases marginally when NTE is deleted (Table 1), particularly in the WT protein. Such changes might arise from structural rearrangements in the barrel (33). We have previously reported that the association affinity of FL WT and C0 to their refolding environment depends on barrel rigidity (31). It is likely that NTE removal alters the barrel scaffold. Taken together with cysteines, we observe that NTE influences the voltage sensing of hVDAC-2, can function only in cis, and plays an important role in the measured barrel characteristics.

NTE + NTH Lowers Barrel Refolding Rates and NTE Minimizes Off-pathway Aggregates in Cys-rich WT—Our electrophysiology measurements establish that the N-helix of hVDAC-2 is crucial for voltage gating and sensing. Generally, residues conferring functionality to proteins also affect other biophysical properties such as (un)folding and stability (41). To address this, we compared the folding/unfolding characteristics of the Δ1–11 and Δ1–32 constructs to the full-length barrels. We chose two 12-C micellar environments for our folding experiments: LDAO (zwitterionic detergent with small headgroup) and DDM (non-ionic detergent with a large headgroup). LDAO forms small spherical micelles (33), whereas DDM micelles are oblate (42), bulky (50–71 kDa), with a higher aggregation number (98-140). Variations in the micelle shape, size, and headgroup can create differences in the hydrophobic radius (42) (see Fig. 1D), ability to

![FIGURE 3. Channel insertion events of hVDAC-2 and its mutants.](image)

**TABLE 1**

Summary of electrophysiological parameters of hVDAC-2 mutants measured in DiPhPC membranes

<table>
<thead>
<tr>
<th>hVDAC-2 mutants*</th>
<th>Positive voltages</th>
<th>Negative voltages</th>
<th>Single channel conductance (nS)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>( n )</td>
<td>( V_0 )</td>
<td>( nFV_0 )</td>
</tr>
<tr>
<td>FL WT (5)</td>
<td>3.12 ± 0.50</td>
<td>28.63 ± 1.92</td>
<td>8.62</td>
</tr>
<tr>
<td>Δ1–31 WT (4)</td>
<td>2.26 ± 0.39</td>
<td>24.15 ± 0.91</td>
<td>5.27</td>
</tr>
<tr>
<td>Δ1–31 WT</td>
<td>2.40 ± 0.66</td>
<td>21.07 ± 1.72</td>
<td>4.88</td>
</tr>
<tr>
<td>FL C0 (6)</td>
<td>2.79 ± 0.17</td>
<td>21.71 ± 1.51</td>
<td>5.84</td>
</tr>
<tr>
<td>Δ1–32 C0</td>
<td>1.84 ± 0.43</td>
<td>32.88 ± 0.17</td>
<td>4.14</td>
</tr>
</tbody>
</table>

* Numbers in parentheses indicate the number of independent experiments conducted for the voltage ramp studies to derive \( n \) and \( V_0 \).

\( nFV_\sigma \) is given in mV.

\( nFV_0' \) is given in kJ mol\(^{-1}\).

**TABLE 1**

Errors in all cases are derived from averaging independent experiments.
incorporate cylindrical or elliptical barrels, and display different protein-micelle interactions. LDAO is structurally analogous to a chemical chaperone (32), and preserves the VDAC structure seen in bilayers (43). DDM is a known stabilizing agent of membrane proteins (44) and maintains the native (functional) protein state.

We used 65 mM LDAO or 19.5 mM DDM for our experiments. These concentrations were chosen to achieve high refolding efficiency, similar micelle number, and comparable secondary structure content for all constructs. The refolding efficiency does not increase further with an increase in the detergent concentration (data not shown). We probed the refolding rates for all the constructs by measuring the intrinsic fluorescence anisotropy (31) of the four interface tryptophans (Fig. 4A). The refolding rates are very fast, and we can capture them only through changes in anisotropy. We see an overall increase in the refolding rates upon deletion of NTH + NTE. In the presence of LDAO, a marginal increase is seen in rates for some constructs, when compared with DDM (Fig. 4B). Our data indicates that in the absence of the N-helix, barrel refolding is kinetically rapid and favorable. NTE + NTH and cysteines together impede refolding. Global comparison of refolding rates of all WT constructs against those for C0 constructs in both LDAO and DDM shows an increase of >20% in the folding rates of the C0 constructs. Hence, our WT constructs display slower folding kinetics than their C0 counterparts (Fig. 4B).

The elevated folding rates of hVDAC-2 Δ1–32 mutants, particularly in LDAO for WT, suggest that the N-helix introduces frustration in the folding pathway for β-barrel formation. NTE + NTH may prevent the formation of native contacts in the barrel or stabilize a refolding intermediate. Hence, why were residues 1–11 retained in the protein? To understand this, we carried out equilibrium refolding measurements of all barrels in LDAO. Interestingly, considerable non-disulfide aggregates form at intermediate GdnHCl concentrations in all proteins (Fig. 4C). Furthermore, Δ1–11 WT and Δ1–32 WT are the only barrel forms to be substantially affected by off-pathway protein aggregation. Such aggregates have been observed for both soluble and membrane proteins (45), and in hVDAC-2, occur at denaturation concentrations that allow the formation of folding intermediate.

Our study shows that in hVDAC-2 WT, NTE (+NTH) lowers the barrel refolding rate to minimize off-pathway protein aggregation. The concomitant absence of aggregates in this system (discussed later), shed valuable light on hVDAC-2 folding mechanism. Cysteines have high hydrophathy values (46), and exposed thiols in the refolding intermediate can augment protein association and result in nonspecific disulfides. Therefore, NTE (+ NTH) of Cys-rich hVDAC-2 can play a crucial role in directing proper barrel formation, by interacting with exposed hydrophobic surfaces and establishing transient disulfides. Our assumption is supported by the altered aggregate characteristics of the refolding intermediate observed between 1 and 3 M GdnHCl in hVDAC-2 C0 (Fig. 4C). We conclude that NTE is important for the proper refolding of hVDAC-2 WT and plays a crucial role in solubilizing the refolding intermediate.

NTE and NTH Impart Thermodynamic Stability to hVDAC-2 in DDM with Structural Rearrangements Evident in LDAO—We next measured the contribution of NTE, NTH, and cysteines to hVDAC-2 thermodynamic behavior, using equilibrium denaturation measurements. We obtain irreversible aggregation of hVDAC-2 upon prolonged incubation in LDAO (Fig. 4C). Furthermore, the refolding process prior to the onset of aggregation is three-state, with a prominent refolding intermediate (Δ1–32 > Δ1–11 > FL for occurrence of the intermediate). However, barrel unfolding can be adequately explained using a two-state model. The resultant hysteresis in the system allows us to estimate only the ΔG^0_app in LDAO (Table 2, discussed later).

The unfolding and refolding profiles for all proteins in DDM show cooperative two-state transitions from the micelle-solvated refolded barrel to the GdnHCl-solvated unfolded state (Fig. 5A). Protein aggregation is considerably diminished and is observed only upon prolonged incubation (>48 h) (data not shown). An appreciable amount of hysteresis is obtained only in
the Δ⁻¹⁻⁻ constructs. This stems from differences in the cooperativity of the refolding versus unfolding processes (low unfolding m value; Fig. 5B). In DDM, NTE deletion causes barrel destabilization by ~0.5 kcal/mol (for WT) and ~0.9 kcal/mol (for C0), whereas obliterating the N-helix further lowers barrel stability by ~1.3 kcal/mol for both proteins (FL versus unfolding ΔG⁰好似 of Δ⁻¹⁻⁻) (Fig. 5B). We have previously observed that in LDAO, FL C0 relies on intra-protein interactions to maintain structural integrity (31). Lowered Cₘ values in FL C0 (Fig. 5B) supports a similar behavior of the Cys-less barrel in DDM. Truncation of the N-helix increases the Cₘ, whereas lowering the unfolding cooperativity, suggesting that intra-protein interactions are replaced with protein-micelle interactions in these proteins. Our data, therefore, underlie the importance of NTE and NTH for thermodynamic stability of hVDAC-2, and connects the functional role to the biophysical property of the protein.

As we observe hysteresis and protein precipitation during hVDAC-2 refolding in LDAO, we extracted information on barrel-micelle interaction strengths from only the unfolding measurements. Although unfolding is a two-state event, the m values in Table 2 are unusually low for an ~32 kDa protein. Such low m values indicate that the unfolding titration corresponds to a hidden three-state process. Therefore, the Gibbs free energy is underestimated in our experiments (47). In line with observations in DDM, the Cₘ of WT constructs are higher than the C0 counterparts (Table 2). The observed increase in ΔG⁰好似 when the N-helix is deleted arises from an increase in both m好似 and Cₘ. In this case, ΔG⁰好似 embodies the solvability of the refolded protein, which, in turn, reflects the protein-micelle interaction strengths. As seen for DDM, removal of the N-helix therefore increases the affinity of the barrel for its lipid milieu.

When compared with DDM, the decreased cooperativity of hVDAC-2 constructs in LDAO shows that the denaturant gains slower access to the barrel in this micelle, thus increasing the transition region of unfolding. To examine if this is due to protein-LDAO interactions, we compared the accessibility and degree of solvent exposure of the four intrinsic tryptophans. The Kₛᵥ (represents fluorophore-quencher interaction kinetics), kₑ (represents quenching efficiency or degree of tryptophan exposure), r (measures the ability of fluorophore to reorient in the excited state), and τₑ (measures the rate of fluorophore rotational diffusion) are provided in Table 3. Kₛᵥ values are lower in DDM compared with LDAO, for both proteins. Similar lifetimes in both micellar systems, however, suggest that the solvent-exposed indoles in LDAO maintain a rigid local environment, whereas the flexible indoles in DDM micelles are occluded from the solvent. This is supported by the higher τₑ in DDM, suggesting that the bulky headgroup of DDM occludes Trp from fluorescence quenchers. Hence, LDAO can form better interactions with hVDAC-2, and facilitates optimal positioning of the indole at the interface. Comparison of the secondary structure content of the refolded proteins using far-UV CD wavelength scans further supports that hVDAC-2 adopts slightly higher β-sheet content in LDAO (compare the MRE₂₃₁₅ values in LDAO and DDM in Table 4).

By and large, in LDAO, Trp accessibility in WT constructs increase when NTE and NTH are deleted, whereas Trp accessibility in C0 counterparts decrease. Within the same micelle system, N-helix and cysteines have complementary effects, promoting barrel-micelle interactions in WT. Furthermore, in LDAO and DDM, lowered Kₛᵥ and τₑ for the WT constructs indicates lipid-solvated indoles, whereas C0 indoles are more solvent accessible. Such local changes in the Trp environment of WT can arise from structural rearrangements driven by changes in the shear number. This has been speculated for VDACs (48). Therefore, a lowered protein-micelle interaction in C0 explains our observed thermodynamic and spectroscopic properties.

### Table 2

Apparent thermodynamic parameters in 13 mM LDAO.

<table>
<thead>
<tr>
<th>Mutants</th>
<th>ΔG₀好似</th>
<th>m好似</th>
<th>Cₘ</th>
</tr>
</thead>
<tbody>
<tr>
<td>FL WT</td>
<td>4.38 ± 0.14</td>
<td>−1.39 ± 0.04</td>
<td>3.15 ± 0.02</td>
</tr>
<tr>
<td>Δ⁻¹⁻⁻ WT</td>
<td>5.76 ± 0.19</td>
<td>−1.62 ± 0.05</td>
<td>3.55 ± 0.01</td>
</tr>
<tr>
<td>Δ⁻¹⁻⁻ C0</td>
<td>6.35 ± 0.24</td>
<td>−1.79 ± 0.06</td>
<td>3.54 ± 0.01</td>
</tr>
<tr>
<td>FL C0</td>
<td>3.21 ± 0.11</td>
<td>−1.13 ± 0.04</td>
<td>2.68 ± 0.02</td>
</tr>
<tr>
<td>Δ⁻¹⁻⁻ C0</td>
<td>2.57 ± 0.08</td>
<td>−1.02 ± 0.03</td>
<td>2.50 ± 0.02</td>
</tr>
<tr>
<td>Δ⁻¹⁻⁻ C0</td>
<td>4.65 ± 0.14</td>
<td>−1.55 ± 0.05</td>
<td>3.00 ± 0.01</td>
</tr>
</tbody>
</table>

![Graph showing chemical denaturation studies of hVDAC-2 in DDM and LDAO.](image-url)
TABLE 3
Tryptophan fluorescence properties of hVDAC-2 and its mutants

<table>
<thead>
<tr>
<th>Micelle</th>
<th>Mutants</th>
<th>hVDAC-2 WT</th>
<th>hVDAC-2 C0</th>
</tr>
</thead>
<tbody>
<tr>
<td>13 mM LDAO</td>
<td>FL</td>
<td>2.50</td>
<td>1.83</td>
</tr>
<tr>
<td></td>
<td>Δ1–11</td>
<td>2.58</td>
<td>1.97</td>
</tr>
<tr>
<td></td>
<td>Δ1–32</td>
<td>2.52</td>
<td>2.09</td>
</tr>
<tr>
<td>3.9 mM DDM</td>
<td>FL</td>
<td>2.50</td>
<td>1.52</td>
</tr>
<tr>
<td></td>
<td>Δ1–11</td>
<td>2.42</td>
<td>1.50</td>
</tr>
<tr>
<td></td>
<td>Δ1–32</td>
<td>2.35</td>
<td>1.55</td>
</tr>
</tbody>
</table>

a <20> given in 10°s.
K_m in M⁻¹.
b kq in 10⁻⁹ mol⁻¹ s⁻¹.
c r in arbitrary units.
d τ_c in 10⁻³.

Thermal Denaturation Measurements Reveal the Role of NTE and NTH in Stabilizing a Compact Barrel—Our previous studies on thermal denaturation of hVDAC-2 revealed that the native barrel is susceptible to heat-mediated irreversible unfolding, which is tightly associated with aggregation (49). A measure of the changes in the secondary structure through far-UV CD (Fig. 6 and Table 4) provides information on scaffold stability and protein-micelle interactions. Assessment of the observed reduction at 215 nm (Fig. 6) suggests a micelle- and N-helix dependence to the aggregation event. The T_m is comparable for all constructs (Table 4). Overall, the values are marginally higher in LDAO than DDM, as expected. Examination of the thermal denaturation-derived parameters for the NTE + NTH deletions reveals that the significant contribution of the N-helix is in stabilizing the barrel scaffold. The T_m,start (35) is lower in most of the constructs lacking NTE or NTH (Table 4). Furthermore, removal of NTE lowers the unfolding cooperativity of hVDAC-2 WT in both LDAO and DDM micelles by ~20 kcal/mol (Table 4). Complete deletion of the N-helix further affects the unfolding cooperativity by ~10 kcal/mol. hVDAC-2 Δ1–32 C0 barrel also exhibits lower unfolding cooperativity by ~15 kcal/mol in DDM and ~30 kcal/mol in LDAO micelles, although the deletion of NTE barely changes the cooperativity in C0 (compare FL C0 and Δ1–11 C0 in Table 4).

This change in cooperativity represents the change in enthalpy during protein unfolding. Refolded hVDAC-2 enthalpy lies in the range of 0.13–0.33 kcal/mol/residue, which is similar to other membrane proteins, but is significantly less than values observed for soluble proteins (50). Low enthalpy for membrane proteins are presumed to stem from a structured denatured state, contributions solely from the extra-membrane segments and/or a buried hydrophobic region upon unfolding (50, 51). In hVDAC-2, barrel unfolding leads to aggregation that buries a considerable region of the protein. These aggregates are not structured (data not shown). Hence, our enthalpy lies in the range of 0.13–0.33 kcal/mol/residue, which is similar to other membrane proteins, but is significantly less than values observed for soluble proteins (50). Low enthalpy for membrane proteins are presumed to stem from a structured denatured state, contributions solely from the extra-membrane segments and/or a buried hydrophobic region upon unfolding (50, 51). In hVDAC-2, barrel unfolding leads to aggregation that buries a considerable region of the protein. These aggregates are not structured (data not shown). Hence, our enthalpy measurements reflect only the change associated with solvent-exposed regions, and mainly involve the N-helix.

Among the FL constructs, C0 exhibits higher enthalpy than WT (Table 4) that suggests a higher exposed surface area in this protein. This observation is in agreement with our fluorescence measurements that C0 shows poor protein-micelle interac-
Molecular Regulators of hVDAC-2 Stability and Function

When the solvent-accessible N-helix is deleted, the enthalpy is lowered. The N-helix of hVDAC-2 is structured (Fig. 2C) and establishes favorable energetic contacts (30), contributing to the high enthalpy of FL constructs. NTE deletion drastically reduces the enthalpy for only the WT barrel (Table 4), convincing us that this segment exists in a solvent-exposed structured state. On the contrary, C0 barrel rigidity and the lack of cysteines can cause the NTE to remain unstructured, as a result of which we see almost no change in enthalpy upon its deletion (Table 4). The NTH is amphipathic, can dock onto the barrel wall (4–6, 9, 10), and lowers the solvent accessibility of these side chains. Hence, the deletion of NTH shows only small enthalpy changes in both WT and C0 (Table 4).

Across our hVDAC-2 deletions, a part of the decrease in enthalpy can also stem from removal of the contacts that the N-helix forms with the transmembrane β-barrel. Another likely explanation is the destabilization of the refolded state of Δ1–11 and Δ1–32 constructs; hence, fewer stabilizing interactions are formed. The latter explanation is also supported by our nFV0 values and chemical denaturation studies. A clear picture that emerges when we consider the unfolding cooperativity, enthalpy, Tm, and Tm-start is that NTE and NTH are critical factors for the formation of a well-structured hVDAC-2. Additionally, from the chemical and thermal denaturation data, we find that NTE adopts a soluble structured conformation in WT hVDAC-2, whereas in C0, it can remain unstructured.

Refolding of hVDAC-2 in High Bicelle q Is Dependent on N-helix, with Micelles and Bicelles Showing Comparable Characteristics—The behavior of membrane proteins in micelles is believed to be different from lipid bilayers. However, in VDACs, studies have affirmed that the barrel structure in LDAO and DMPC are similar (43). Nonetheless, whereas micelles are dynamic entities, lipid bilayers exert transverse and lateral forces, due to hydrophobic mismatch and packing pressure at the membrane–protein junction. To assess the requirement of NTE and NTH for hVDAC-2 in membrane, we used membrane bicelles of various q.

We monitored hVDAC-2 behavior using CD, Tm, and accessibility of tryptophans (KSv) (Fig. 7) (31). We obtained comparable refolding efficiency in all bicelle q from 0.0015 to 1.0, except for Δ1–32 WT, which failed to refold optimally at q > 0.3 in DMPC:LDAO (20–30% higher unfolded fraction (fU) in Fig. 7, A and D). Microcalorimetry measurements (Fig. 7, G and H) reveal that at q > 0.3 (DMPC:LDAO) or q > 1 (DMPC:DDM) the bicelle phase transition changes from small isotropic micelles (at low q) to bilayer-like character. Unassisted refolding of hVDAC-2 is, therefore, lower in membrane systems, and displays the order C0 > WT and FL > Δ1–11 > Δ1–32. NTE and NTH are therefore required for the hVDAC-2 barrel to refold in lipidic systems.

DDM supports hVDAC-2 refolding (compare Fig. 7, A and D) in micelles of higher q, possibly because of its oblate micellar shape. Stronger protein–bicelle interactions in DMPC-DDM micelles may also be responsible for this. Lower KSv for the WT barrels in DMPC:LDAO and DMPC:DDM, as compared with the C0 constructs (Fig. 7, B and E, left versus right panels) suggests that indoles of the WT constructs are more lipid solvated. Hence, the strong barrel-micelle interaction exerted by the WT protein is also translated to lipid bicelles. The truncation mutants display low KSv in conditions with poorer barrel refolding efficiency (Fig. 7, B and E). When we consider this data with the observed loss in secondary structure (f0 values in Fig. 7, A and D), the formation of a membrane-adsorbed barrel upon N-helix deletion is evident. As seen in micelles (Fig. 4B),...
NTE + NTH facilitate proper hVDAC-2 refolding in lipid bicelles.

Thermal stability measurements (Fig. 7, C and F) provide us with several interesting facets of hVDAC-2 behavior. (i) Similar $T_m$ values for all constructs across both bicelle systems suggest that our measurements correspond primarily to protein-DMPc interactions. (ii) The $T_m$ measured in micelles ($q = 0$) is retained in bicelles ($q \leq 0.015$ for DMPc:LDAO and $q \leq 0.05$ for DMPc:DDM) (Fig. 7, C and F). This indicates a structural similarity of hVDAC-2 between micelles and lipids, as seen for hVDAC-1 (43). (iii) In higher $q$, where the barrel exists in the adsorbed state, both WT and C0 barrels show comparable $T_m$, although $f_{1C0}$-WT $< f_{1C0}$-C0 (compare Fig. 7, A and C or D and F). The adsorbed state of C0 constructs are, therefore, more $\beta$-rich. This is in good agreement with our previous observation that Cys-less constructs are well structured (32, 49). (iv) In DMPc: DDM bicelles, both the $\Delta^{1-32}$ constructs display a marginally higher $T_m$ value (Fig. 7F), although the $f_1$, and $K_{SV}$ values are similar to the FL and $\Delta^{1-11}$ barrels. The NTE + NTH-deleted hVDAC-2 may possibly tolerate higher lateral bilayer pressure in DMPc:DDM bicelles.

Our results clearly demonstrate that NTE and NTH are required for proper refolding of hVDAC-2 in lipidic systems. Once the correctly folded form is achieved, the stability of the barrel is primarily determined by the transmembrane $\beta$-sheets that form the pore. Most interestingly, the thermal stability dramatically decreases at high bicelle $q$ (Fig. 7, C and F), suggesting that unlike hVDAC-1, hVDAC-2 requires additional stabilizing factors to sustain the bilayer pressure. In the mitochondrial membrane, these may include barrel oligomerization, interaction with lipids like cardiolipin or cholesterol, and binding to anti-apoptotic agents such as BAK.

**Discussion**

The N-terminal helix of VDAC has attracted recent attention, due to its requirement for channel function and its involvement in several regulatory processes, interaction with anti-apoptotic proteins, and cell survival (19, 20, 29). Additionally, the unique 11-residue sequence observed only in human isofrom 2 is of intrigue from the structural and functional perspectives. However, studies on this isoform have been surprisingly limited. Our report is the first to embody the functional and biophysical requirement of the unique 11-residue NTE in hVDAC-2. We find that NTE is structured and plays an important role in the proper refolding of hVDAC-2 in both micelles and lipid bilayers. NTE suppresses the formation of undesirable off-pathway aggregates, and stabilizes the native state. Functionally, NTE confers asymmetric gating to the hVDAC-2 channel in DiPhPC membranes.

NTE exerts an important influence on the barrel characteristics in the presence of cysteines. It is well known that the most abundant VDAC-1 isoform possesses only two cysteines; these cysteines are not important for apoptosis (52). However, hVDAC-2 has nine cysteines that orient toward the intermembrane space. Considering the sparsity of cysteines in proteins, the cysteine content of hVDAC-2 is unusually high. We have previously demonstrated that when cysteines are mutated, hVDAC-2 can exhibit characteristics akin to hVDAC-1 (31). NTE deletion in this Cys-less construct shows a less pronounced functional and biophysical manifestation compared with NTE-deleted hVDAC-2 WT. Our experimental observations allow us to conclude that the Cys-enriched WT barrel has specifically retained the 11-residue NTE for stability and function. Our hypothesis is additionally supported by the conservation of cysteines in VDAC-2, particularly in higher mammals that also possess NTE (Fig. 8 and supplemental Fig. S1). We are tempted to consider that in the course of evolution, VDAC-2 sequences of higher mammals, whereas gaining cysteines, may also have gained NTE to stabilize the barrel and as a means of incorporating more cysteines. Only two conserved cysteines are retained in all three isoforms and across most organisms (supplemental Figs. S1–S3), and these VDACs are devoid of NTE. NTE is therefore functionally unimportant for hVDAC-1, whereas it is evolutionarily significant in hVDAC-2.

VDACs are not the only known transmembrane $\beta$-barrels with a biologically important N-helix. The bacterial outer membrane enzyme PagP, for example, anchors itself to the phospholipid membrane through the N-terminal helix (53, 54). hVDAC-2 NTE(+NTH) serves the additional role of assisting proper refolding of the protein and in voltage sensing. Previous studies on hVDAC-1 have proposed a loss in barrel topology upon channel closure and deletion of NTH (7, 10, 48). The elliptical topology of the closed channel, seen upon removal of the N-helix, shows altered strand tilting (7, 48). Our experiments also demonstrate that externally supplemented NTH is unable to revive the voltage gating of hVDAC-2 $\Delta^{1-32}$. Could a similar elliptical topology therefore be accessible to hVDAC-2? The differential barrel-lipid interactions and structure-functional studies we observe suggest that $\Delta^{1-32}$ C0 can access an elliptical closed state. In WT, on the other hand, our biophysical results support strong barrel-lipid interactions. Particularly, the observation that $\Delta^{1-32}$ WT mainly inserts in the open state as opposed to $\Delta^{1-32}$ C0 barrel, strongly advocates for a less collapsed topology in native hVDAC-2. Another interesting outcome of our study is the influence of cysteines on the barrel scaffold. Cysteines affect hVDAC-2 folding and the barrel tolerance to increased bilayer pressure when NTE + NTH are absent. Remarkably, the barrel-lipid interaction strengths are altered in the $\Delta^{1-32}$ C0 barrel, which could stem from the elliptical conformation. Such interactions depend on whether the micelle packing can accommodate the altered topology. LDAO experiences electrostatic repulsion in the headgroup (55) and are more suited to stabilize spherical hVDAC-2 structures that impose lesser demands on micelle packing. The non-ionic oblate DDM micelles have lesser hydrophobic diameter and do not suffer from packing defects as LDAO (55). DDM can therefore promote protein-micelle interactions in both spherical and elliptical barrel forms. This feature of DDM is possibly why we obtain DMPc:DDM bicelles of higher $q$ and optimal lateral pressure for refolding hVDAC-2.

In hVDAC-2, our experiments in 12-C micellar systems translate reproducibly to the 14-C lipidic bicelles. A similar structural scaffold is therefore retained by this protein in various systems. Furthermore, the contribution of NTE and NTH are comparable in LDAO, DDM and DMPc. Our measurements, therefore, correspond primarily to the interaction of the
N-helix with the β-barrel, and the β-barrel with its lipid environment. The intrinsic amino acid composition, secondary structure, and tertiary interactions mediated through NTH and NTE together contribute substantially to the overall hVDAC-2 stability. hVDAC-2 topology and strand tilt can be altered by the lipid environment (7, 48), in a cysteine- and N-helix dependent manner, through likely changes in the shear number. The loss in barrel stability with an increase in bilayer lateral pressure suggests that agents such as unsaturated lipids, which introduce defects in ordered lipids, promote refolding and control residence time of the barrel in the mitochondrial outer membrane.

Our observation that hVDAC-2 can remain in the “open” state even in the absence of the N-helix opens new avenues for its contribution to apoptosis. The role of cysteines of hVDAC-2 and hVDAC-3 in ROS regulation has been speculated for some time. Irreversible cysteine oxidation during ROS build-up can undock the N-helix from within the barrel. As hVDAC-2 can retain its open state in this condition, it is tempting to speculate that this conformation can ultimately lead to unregulated mitochondrial outer membrane permeabilization and cell death. Furthermore, conformational changes imposed by lateral bilayer pressure and N-helix interaction can cause structural changes and lead to different barrel stability. As has been recently seen for BAX-mediated apoptosis (56), this process can trigger or halt the apoptotic pathway.

Author Contributions—R. M. conceived the study. S. R. M. performed the experiments. Both authors analyzed the data and wrote the paper.

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