

Structure and Kinetics of Monofunctional Proline Dehydrogenase from *Thermus thermophilus**[§]

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Tommi A. White[‡], Navasona Krishnan[§], Donald F. Becker[§], and John J. Tanner^{†¶1}

From the Departments of [‡]Biochemistry and [¶]Chemistry, University of Missouri-Columbia, Columbia, Missouri 65211 and the [§]Department of Biochemistry, Redox Biology Center, University of Nebraska, Lincoln, Nebraska 68588

Proline dehydrogenase (PRODH) and Δ^1 -pyrroline-5-carboxylate dehydrogenase (P5CDH) catalyze the two-step oxidation of proline to glutamate. They are distinct monofunctional enzymes in all eukaryotes and some bacteria but are fused into bifunctional enzymes known as proline utilization A (PutA) in other bacteria. Here we report the first structure and biochemical data for a monofunctional PRODH. The 2.0-Å resolution structure of *Thermus thermophilus* PRODH reveals a distorted $(\beta\alpha)_8$ barrel catalytic core domain and a hydrophobic α -helical domain located above the carboxyl-terminal ends of the strands of the barrel. Although the catalytic core is similar to that of the PutA PRODH domain, the FAD conformation of *T. thermophilus* PRODH is remarkably different and likely reflects unique requirements for membrane association and communication with P5CDH. Also, the FAD of *T. thermophilus* PRODH is highly solvent-exposed compared with PutA due to a 4-Å shift of helix 8. Structure-based sequence analysis of the PutA/PRODH family led us to identify nine conserved motifs involved in cofactor and substrate recognition. Biochemical studies show that the midpoint potential of the FAD is -75 mV and the kinetic parameters for proline are $K_m = 27$ mM and $k_{cat} = 13$ s⁻¹. 3,4-Dehydro-L-proline was found to be an efficient substrate, and L-tetrahydro-2-furoic acid is a competitive inhibitor ($K_I = 1.0$ mM). Finally, we demonstrate that *T. thermophilus* PRODH reacts with O₂ producing superoxide. This is significant because superoxide production underlies the role of human PRODH in p53-mediated apoptosis, implying commonalities between eukaryotic and bacterial monofunctional PRODHs.

Oxidation of amino acids is a central part of energy metabolism. The oxidative pathway for proline consists of two enzymatic steps and an intervening nonenzymatic equilibrium (Scheme 1) (1, 2). The first enzymatic step transforms proline to Δ^1 -pyrroline-5-carboxylate (P5C),² which is non-enzymatically hydrolyzed to glutamic semialdehyde. The semialdehyde is oxidized in the second enzymatic step to glutamate.

This 4-electron transformation of proline is common to all organisms, but the enzymes of proline catabolism differ widely among the three kingdoms of life. Amino acid sequence analysis shows that bacteria and eukaryotes share a common set of proline catabolic enzymes called proline dehydrogenase (PRODH) and P5C dehydrogenase (P5CDH). Studies of the bacterial enzymes have shown that PRODH is an FAD-dependent enzyme with a $(\beta\alpha)_8$ barrel catalytic core (3, 4), and P5CDH is an NAD⁺-dependent Rossmann fold enzyme featuring a nucleophilic Cys (5). These enzymes are unrelated in sequence and structure to hyperthermophilic archaeal proline catabolic enzymes, which appear in unique hetero-tetrameric and hetero-octameric complexes (6).

An intriguing aspect of proline catabolism in eukaryotes and bacteria is that PRODH and P5CDH are separate enzymes in some organisms, whereas the two enzymes are fused in other organisms. The traditional view has been that PRODH and P5CDH appear as separate enzymes in eukaryotes and fused bifunctional enzymes known as proline utilization A (PutA) (7–15) in bacteria. Fusion of enzymes catalyzing sequential steps of a metabolic pathway provides a kinetic advantage, because the intermediate can be channeled between active sites (16, 17). Indeed, Maloy's group reported kinetic data supporting substrate channeling for *Salmonella typhimurium* PutA (18). Eisenberg and coworkers (19) refer to fused proteins, such as PutA, as Rosetta Stone proteins, because they decipher interactions between protein pairs. Thus, the Rosetta Stone hypothesis of protein evolution predicts that eukaryotic PRODH and P5CDH form physical and functional interactions.

As we elaborate in this report, separate PRODH and P5CDH enzymes are not restricted to eukaryotes, but also appear in

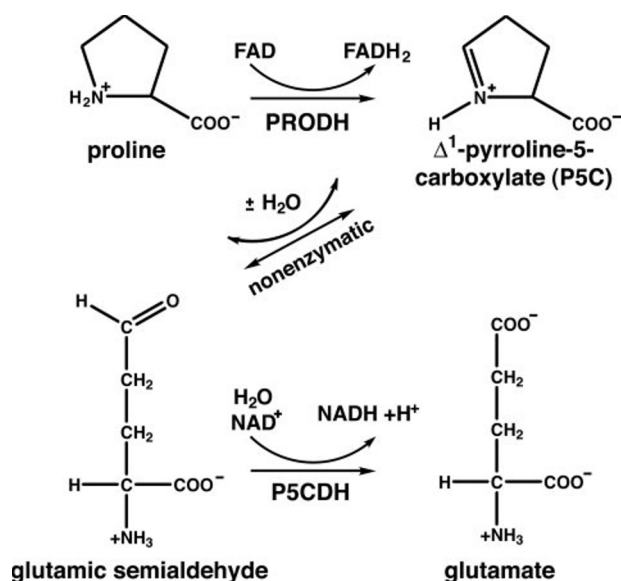
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The atomic coordinates and structure factors (code 2G37) have been deposited in the Protein Data Bank, Research Collaboratory for Structural Bioinformatics, Rutgers University, New Brunswick, NJ (<http://www.rcsb.org/>).

[§] The on-line version of this article (available at <http://www.jbc.org>) contains supplemental Fig. S1.

¹ To whom correspondence should be addressed: Dept. of Chemistry, University of Missouri-Columbia, Columbia, MO 65211. Tel.: 573-884-1280; Fax: 573-882-2754; E-mail: tannerjj@missouri.edu.

² The abbreviations used are: P5C, Δ^1 -pyrroline-5-carboxylate; PRODH, proline dehydrogenase; P5CDH, Δ^1 -pyrroline-5-carboxylate dehydrogenase; PutA, proline utilization A; TtPRODH, *T. thermophilus* proline dehydrogenase; TtP5CDH, *T. thermophilus* Δ^1 -pyrroline-5-carboxylate dehydrogenase; DCPIP, dichlorophenolindophenol; THFA, L-tetrahydro-2-furoic acid; o-AB, o-aminobenzaldehyde; TIM, triosephosphate isomerase; SASA, solvent-accessible surface area; ROS, reactive oxygen species; PDH1 and PDH2, proline dehydrogenase complexes from *Pyrococcus horikoshii*; DTT, dithiothreitol; MPD, 2-methyl-2,4-pentanediol.



some bacteria. These bacterial monofunctional enzymes thus represent a new and unexplored group of enzymes. In addition, bacterial monofunctional PRODHs may serve as good models for understanding human PRODH. Moreover, the bacterial enzymes are convenient systems for studying protein-protein interactions and intermolecular channeling. To begin exploring these questions, we previously reported the cloning, isolation and crystallization of the monofunctional PRODH from *Thermus thermophilus* (TtPRODH) (20). Here we report the crystal structure and kinetic characterization of this enzyme.

EXPERIMENTAL PROCEDURES

Expression, Purification, and Crystallization of Se-Met TtPRODH—A 10-ml overnight culture grown in LB media was pelleted, and the LB medium was removed. The cells were resuspended in 3 ml of M9 media and diluted into 1.5 liters of M9 media. The culture was then grown to an optical density of 0.5 ($\lambda = 600$ nm). Methionine production was inhibited for 30 min as described previously (21) followed by induction of protein expression by the addition of 0.5 mM isopropyl 1-thio- β -D-galactopyranoside. After 12 h of induction at 22 °C with a shaking rate of 200 rpm, cells were harvested, resuspended in 50 mM NaH_2PO_4 at pH 8.0, 300 mM NaCl, 10 mM imidazole, 1 mM DTT, and 5% glycerol, and frozen.

Se-Met TtPRODH was purified using procedures described for purification of TtPRODH (20), except for the following modifications. As observed for native TtPRODH, the pellet obtained after centrifugation of lysed cells was bright yellow, indicating that a significant amount of Se-Met TtPRODH was bound to the cell debris in addition to Se-Met TtPRODH in the supernatant. Therefore, the enzyme was extracted from the pellet by resuspending the pellet in 20 ml of 20 mM *n*-octyl- β -D-glucopyranoside, 50 mM NaH_2PO_4 , 300 mM NaCl, 10 mM imidazole, 1 mM DTT, 5% glycerol, pH 8.0, followed by centrifugation. The detergent extraction step was repeated once or twice until no additional TtPRODH was liberated. The extracted protein was then purified using nickel-nitrilotriacetic acid chromatography as described previously (20).

Purified Se-Met TtPRODH was dialyzed overnight in the dark into 4 liters of 50 mM Tris-HCl, pH 8.0, 50 mM NaCl, 0.5 mM EDTA, 5 mM DTT, 0.1 mM FAD, and 5% glycerol using 10,000 molecular weight cut-off dialysis tubing (Spectrum Spectra/Por). After dialysis, the enzyme was passed through a 12-ml Sephadex G-25 desalting column (GE Healthcare) to remove excess FAD and then concentrated to 13 mg/ml using centrifugal concentrating devices (Amicon, Millipore). Protein concentration was estimated using the Bradford method (Pierce Coomassie Plus). Molecular masses of Se-Met and native TtPRODH obtained from matrix-assisted laser desorption/ionization time-of-flight mass spectrometry were 38208 ± 6 and 37968 ± 3 Da, respectively, which indicated 100% incorporation of Se-Met into all five Met positions of the TtPRODH polypeptide chain.

Se-Met TtPRODH was crystallized using procedures described previously for native TtPRODH (20). Briefly, crystals were grown at room temperature in sitting drops by mixing equal volumes of the enzyme (2 μ l) and reservoir (2 μ l) solutions. The enzyme solution contained 2–3 mg/ml Se-Met TtPRODH and 20 mM of fresh *n*-octyl- β -D-thioglucopyranoside. The reservoir solution consisted of 100 mM imidazole (pH 7), 100 mM MgCl_2 , 17% MPD, and 5 mM DTT. The crystals typically appeared within 1 day after setup and grew to a maximum dimension of 0.2 mm. Crystals were prepared for cryogenic data collection by soaking in 100 mM imidazole (pH 7), 200 mM MgCl_2 , and 25% MPD, picked up with Hampton mounting loops, and plunged into liquid nitrogen.

The space group is $P2_12_12_1$ with unit cell dimensions of $a = 82.1$ Å, $b = 89.6$ Å, and $c = 94.3$ Å. There are two protein molecules per asymmetric unit, with 46% solvent content and Matthews coefficient of $2.3 \text{ Å}^3 \text{ Da}^{-1}$ (22).

X-ray Diffraction Data Collection, Phasing, and Refinement—Data were collected at Advanced Light Source beamline 4.2.2 using a NOIR-1 charge-coupled device detector. The structure was solved using single-wavelength anomalous diffraction phasing using data collected at the energy corresponding to the experimentally determined maximum of f'' . The data set used for structure determination and refinement consisted of 180 frames with a crystal-to-detector distance of 170 mm, oscillation range of $1^\circ/\text{frame}$, and exposure time of 30 s/frame. Integration and scaling were performed with d*TREK (23). Data collection and processing statistics are listed below in Table 1.

SOLVE (24) was used to identify a constellation of anomalous scattering centers, and the resulting single-wavelength anomalous diffraction phases were improved with solvent flattening in RESOLVE (24). The partial chain trace from RESOLVE was used to determine the non-crystallographic symmetry transformation relating the two molecules in the asymmetric unit. The RESOLVE phases were then improved with non-crystallographic symmetry averaging and solvent flattening in DM (25). The DM phases were input to ARP/wARP for automated model building (26). The model from ARP/wARP was improved with several rounds of model building in COOT followed by refinement against the Se-Met peak data set with REFMAC5 (27). Topology and parameter files for FAD were created using PRODRG (28) and the Libcheck module of CCP4i (29, 30).

Structure of *T. thermophilus* Proline Dehydrogenase

The final model includes residues 5–296 of TtPRODH chain A and residues 1–294 of TtPRODH chain B. In addition, 6 residues of the N-terminal affinity tag were built for chain B (residues –5 to 0). The carboxyl-terminal ends of the protein are disordered with residues 297–307 and 295–307 omitted in chains A and B, respectively. The model also includes 1 FAD cofactor per TtPRODH chain, 272 water molecules, and 4 MPD molecules. See Table 1 for refinement statistics. Coordinates and structure factor amplitudes have been deposited in the Protein Data Bank under accession number 2G37 (31).

PRODH Kinetic Characterization—TtPRODH was expressed in *Escherichia coli* and purified as described previously (20). The purified enzyme was dialyzed into buffer containing 50 mM Tris-HCl, 50 mM NaCl, 0.5 mM EDTA, 0.5 mM DTT, and 5% glycerol at pH 8.0 and stored at 4 °C. PRODH activity was measured using the proline-dichlorophenolindophenol (DCPIP) oxidoreductase assay as described previously for PutAs (11). One unit of PRODH activity is the quantity of enzyme that transfers electrons from 1 μ mol of proline to DCPIP per minute at 25 °C. Steady-state kinetic parameters for L-proline were obtained using the DCPIP assay with proline as the variable substrate in the range of 0.1–100 mM. Three trials were performed for each proline concentration. The parameters K_m and V_{max} were obtained by fitting the data to the Michaelis-Menten equation using Origin software. Kinetic constants for an alternative substrate, 3,4-dehydro-L-proline, were determined similarly, with the substrate concentration varied in the range 3–500 μ M. Inhibition by L-tetrahydro-2-furoic acid (THFA), L-lactate, and L-mandelate were examined by steady-state inhibition kinetic measurements using proline as the variable substrate. Inhibition data were analyzed using the method of Dixon (32).

Thermostability was assessed by measuring PRODH activity as a function of incubation time at 90 °C. For this study, the enzyme was incubated in a water bath at 90 °C, and aliquots were removed at various time points and stored at 4 °C. After all of the aliquots were taken and cooled to 4 °C, activity assays were conducted at 25 °C in the presence of 25 mM proline.

UV-visible Spectroscopy—Potentiometric titrations of TtPRODH were recorded at 20 °C in 50 mM potassium phosphate buffer (pH 7.5) containing 50 mM NaCl and 5% glycerol using a three-electrode single compartment spectroelectrochemical cell as previously described (14, 33). Measurements were made under a nitrogen atmosphere in a Belle Technology glove box. All potential values are reported *versus* the normal hydrogen electrode. Methyl viologen (20 μ M) was used as a mediator dye, and pyocyanine (5 μ M) and indigo disulfonate (3 μ M) were used as indicator dyes. The UV-visible spectra in each experiment were recorded for $\lambda = 300$ –700 nm on a Cary 100 spectrophotometer. Clean spectra of TtPRODH were obtained by subtracting the spectra of the dyes in the absence of protein measured under identical conditions. The reduction potential (E_m) and n values were calculated by linear regression of a Nernst plot of the data.

TtPRODH was also titrated with proline under aerobic conditions in the proline concentration range 0–160 mM. At each proline concentration, the absorbance spectrum was acquired after incubating the enzyme (56 μ M) for 5 min with

TABLE 1

Data collection and refinement statistics

Values for the outer resolution shell of data are given in parenthesis.

Data collection	
Wavelength (Å)	0.97932
Diffraction resolution (Å)	47.16–2.00
No. of observations	292,071
No. of unique reflections	46,775
Redundancy	6.24 (3.12)
Completeness (%)	98.0 (86.3)
R_{merge}	0.079 (0.399)
Average I/ σ	13.6 (3.0)
Wilson b-factor (Å ²)	16
PDB accession code	2G37
No. of protein chains	2
No. of protein residues	592
No. of protein atoms	5,184
No. of water molecules	272
Refinement statistics	
R_{cryst}	0.190 (0.242)
R_{free}^a	0.231 (0.362)
r.m.s.d. ^b	
Bond lengths (Å)	0.014
Bond angles (deg.)	1.5
Ramachandran plot ^c	
Favored (%)	94.4
Allowed (%)	5.4
Generously allowed (%)	0.2
Average b-factors (Å ²)	
Protein	16
Water	16
FAD	17

^a 5% random R_{free} test set.

^b Compared to the Engh and Huber parameters (57).

^c The Ramachandran plot was generated with PROCHECK (58).

proline. The absorbance values at $\lambda = 450$ nm were analyzed as previously described (11, 34) to obtain an apparent equilibrium constant for the formation of a reduced enzyme-P5C complex (Reaction 1).



REACTION 1

Proline:O₂ Reactivity and Generation of ROS—The rate of proline:O₂ activity was determined by monitoring the appearance of P5C as a function of proline concentration in the absence of an artificial electron acceptor. P5C was detected as the yellow dihydroquinazolinium complex with *o*-aminobenzaldehyde (*o*-AB) by monitoring absorbance at $\lambda = 443$ nm ($\epsilon = 2900 \text{ M}^{-1}\text{cm}^{-1}$) (35). The assay mixture included 50 mM potassium phosphate buffer (pH 7.5), 20 μ g of TtPRODH, 4 mM *o*-AB, and 0.5–25 mM proline. The reaction was monitored for 15 min, and data from the 5- to 10-min time period were used for rate calculations. One unit of proline:O₂ activity is defined as the amount of TtPRODH that generates 1 μ mol of P5C per minute at 25 °C.

Production of superoxide by TtPRODH was studied by measuring reduction of cytochrome *c* as described previously (36). Reduction of cytochrome *c* was indicated by an increase in absorbance at $\lambda = 550$ nm. For these assays, the L-proline concentration was 50 mM, and the concentrations of cytochrome *c* and TtPRODH were 0.02 mg/ml each. Production of H₂O₂ by TtPRODH (0.02 mg/ml) was measured using the Amplex Red H₂O₂/peroxidase assay kit (Molecular Probes, Invitrogen) and a standard curve obtained from solutions of known H₂O₂ concentration.

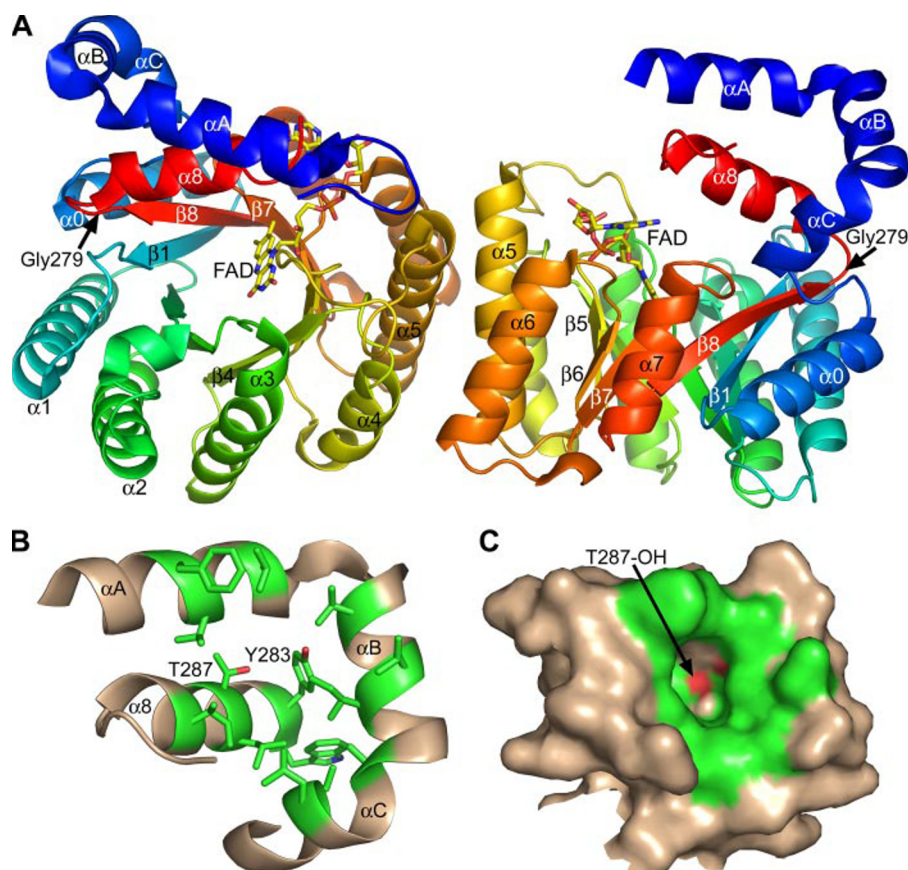


FIGURE 1. **Overall structure of TtPRODHD.** A, ribbon drawing of the two TtPRODHD molecules in the asymmetric unit. The protein chains are colored in the rainbow scheme, with dark blue at the amino terminus and red at the carboxyl terminus. Selected α -helices and β -strands are labeled. The FADs are drawn as stick models in yellow. The glycine hinge between $\beta 8$ and $\alpha 8$ is noted (Gly²⁷⁹). B, the solvent-exposed hydrophobic patch formed by α -helices A, B, C, and 8. The orientation is similar to that of the right-hand protein in A. Side chains of the patch are colored green. C, surface representation of the hydrophobic patch. The orientation is identical to that of B. This figure and others were created with PyMOL (W. L. DeLano (2002) The PyMOL Molecular Graphics System).

Bioinformatics Analysis of the Bacterial PutA/PRODHD Family—Multiple sequence alignment calculations of PutA PRODHD domains and bacterial monofunctional PRODHDs were performed with ClustalW (37). The data set of sequences used for these calculations was obtained from the Protein Information Resource using the Related Sequences tool with TtPRODHD as the query sequence. A total of 287 homologs of TtPRODHD were identified.

RESULTS

Overall Fold—The TtPRODHD structure was determined at 2.0-Å resolution using Se-Met single-wavelength anomalous diffraction phasing (Table 1). There are two protein chains in the asymmetric unit, labeled A and B, and 1 FAD cofactor bound to each protein (Fig. 1A). The structure reveals a distorted ($\beta\alpha$)₈ barrel fold. The first strand of the barrel ($\beta 1$) is preceded by 4 α -helices denoted αA , αB , αC , and $\alpha 0$ (Fig. 1A, the right side provides the best view). The major distortion of the TtPRODHD barrel from the classic triosephosphate isomerase (TIM) barrel concerns the placement of $\alpha 8$. Helix $\alpha 0$ packs against $\beta 1$ thus occupying the location reserved for $\alpha 8$ in the classic TIM barrel. Consequently, $\alpha 8$ of TtPRODHD is located above the carboxyl-terminal face of the β -barrel rather than alongside the β -barrel as in the classic TIM structure.

The FAD cofactor of TtPRODHD is bound at the carboxyl ends of the strands of the barrel (Fig. 1A). The *re* face of the isoalloxazine packs tightly against strands 4–6, whereas the *si* face opens to the substrate binding pocket and is available for hydride transfer from the substrate proline.

Helices A–C and 8 form a small domain above the carboxyl-terminal face of the β -barrel (Fig. 1A, the right side provides the best view). Helices A and B form a right angle, as do helices B and C. Helix 8 fits into the cleft formed by helices A–C. These four helices pack together to form a solvent-exposed hydrophobic patch (Fig. 1, B and C). The patch has a pronounced hole in the middle (Fig. 1C). Several hydrophobic side chains form the rim of the hole and line its sides, whereas the hydroxyl groups of Thr²⁸⁷ and Tyr²⁸³ form the bottom of the hole (Fig. 1, B and C). If TtPRODHD interacts with *T. thermophilus* P5CDH (TtP5CDH), then the function of the hydrophobic patch may be to serve as a docking interface for TtP5CDH. The patch presents a concave hydrophobic surface to solvent (Fig. 1C), and thus it is tempting to speculate that this patch

mates with a complementary convex surface of TtP5CDH. Involvement of this patch in interaction with TtP5CDH also makes sense for substrate channeling, because the patch is close to the TtPRODHD active site.

The pair of molecules chosen for the asymmetric unit corresponds to the largest interface in the crystal lattice, which buries 992 Å² of surface area. The two molecules interact primarily through the packing of $\alpha 5$ (residues 200–218) of one chain against $\alpha 5$ and $\alpha 6$ (residues 228–242) of the other chain (Fig. 1A). The orientation of the two molecules is such that the $\alpha 5$ helices form an angle of 50°. The interface has four intermolecular ion pairs (Lys²¹³–Glu²⁰⁷ and Arg²³⁰–Asp²⁰⁰) and four intermolecular hydrogen bonds (His²¹⁰–His²¹⁰ and Tyr²³⁸–Glu¹⁷⁰). Several hydrophobic side chains are also buried in the interface: Phe¹⁹⁸, Leu²⁰³, Leu²⁰⁹, and Leu²¹⁴.

The significance of this interface is unclear at this time. It is not the one typically found in dimeric TIM, nor is it similar to the tetramer interface of archaeal TIMs (38). The classic TIM dimer interface involves loops 1–4 (38) and buries ~1500 Å² of surface area, which is 50% larger than the TtPRODHD interface. Analysis of detergent-solubilized TtPRODHD with dynamic light scattering and gel filtration suggests the presence of apparent monomeric and dimeric species. It is possible that the pair of proteins in the asymmetric unit represents the dimeric spe-

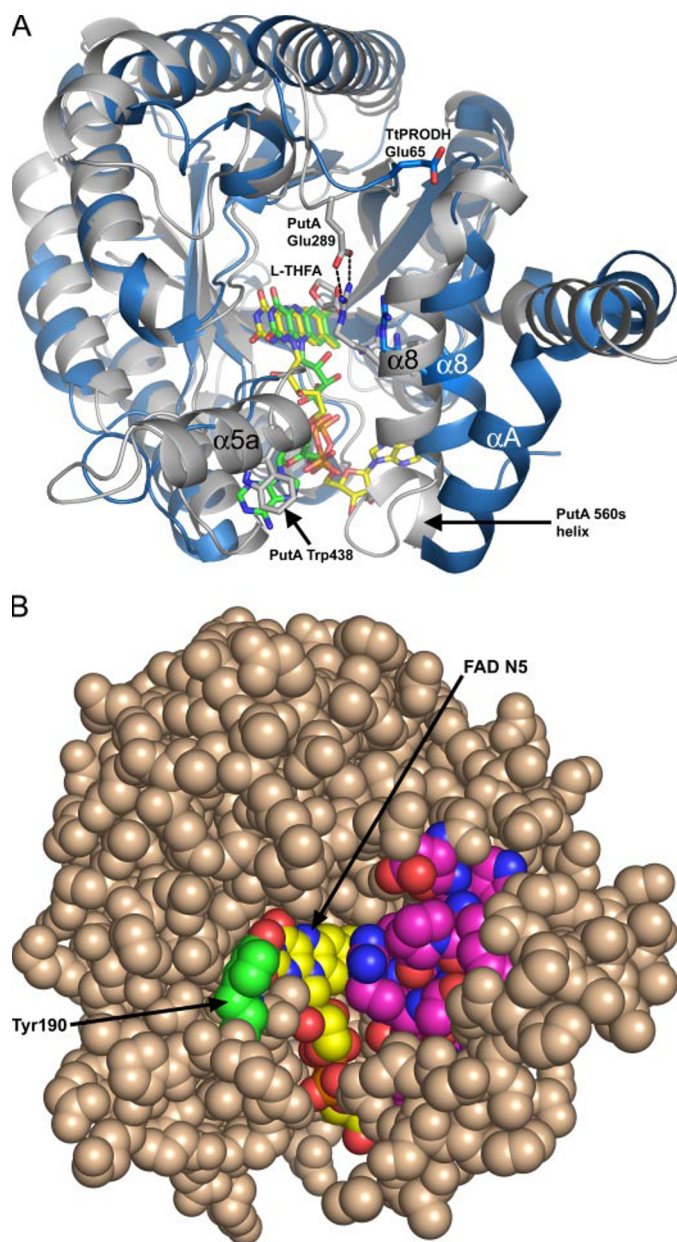


FIGURE 2. **Comparison of TtPRODHD and PutA86-669 complexed with THFA.** A, superposition of TtPRODHD (blue) and PutA86-669/THFA (white, PDB code 1TIW). The FAD cofactors of TtPRODHD and PutA86-669 are yellow and green, respectively. Glu⁶⁵ of TtPRODHD is drawn in stick mode. Trp⁴³⁸ of PutA86-669, which stacks against the FAD adenine, is also drawn. The dashed lines indicate the Glu²⁸⁹-Arg⁵⁵⁵ ion pair in PutA86-669. B, space-filling representation of TtPRODHD emphasizing solvent exposure of the FAD (yellow). Helix $\alpha 8$ is magenta. Tyr¹⁹⁰ is shown in green. The orientation is similar to that of A.

cies observed in solution; however, it is also possible that this interface is an artifact of crystallization. An additional complicating factor in assessing the biological relevance of this interface is the possibility that TtPRODHD interacts with TtP5CDH.

The TtPRODHD structure is only the second PRODH structure solved to date, with the first being the PRODH domain of *E. coli* PutA (PutA86-669, PDB code 1TIW) (3, 4). Residues 263-561 of PutA form a distorted TIM barrel that is similar to that of TtPRODHD (Fig. 2A). Within this catalytic core, the sequence identity between the two enzymes is 29%, and the root

mean square deviation for C $_{\alpha}$ atoms is 1.7 Å for 243 aligned residues. Also, the general location of the FAD at the carboxyl-terminal tips of the strands is similar in the two structures.

The major topological difference between the TtPRODHD and the PutA86-669 barrels is an extra α -helix that appears in PutA86-669 but is absent in TtPRODHD. Residues 437-449 of PutA form an α -helix ($\alpha 5a$ in Fig. 2A) that is replaced by a 5-residue loop in TtPRODHD (residues 190-195). As discussed in the next section, this topological difference has a dramatic effect on FAD conformation.

FAD Conformation—Over a dozen residues interact with the FAD either through electrostatic or non-polar interactions. Electrostatic interactions with the FAD are shown in Fig. 3. The five hydrogen bonds to the isoalloxazine in TtPRODHD are also present in PutA86-669 structures. Accordingly, the positioning of the isoalloxazine within the barrel is similar in TtPRODHD and PutA86-669.

Surprisingly, the FAD conformations of the two enzymes are dramatically different (Fig. 4). In TtPRODHD the adenosine is under the dimethyl benzene ring of the isoalloxazine (Fig. 4A), whereas in PutA86-669 the adenosine is below the pyrimidine ring of the isoalloxazine (Fig. 4B). The difference is dramatic. When superimposed, the adenine groups of the two cofactors are separated by 13 Å (Fig. 4B).

Accordingly, the interactions involving the ribityl, pyrophosphate, and adenosine groups are different in the two enzymes. For example, in TtPRODHD the ribityl 2'-OH is tucked under the pyrimidine ring of the isoalloxazine so that it forms hydrogen bonds with N1 of FAD (2.7 Å, Fig. 4B) and the amine of Gly¹⁸⁸ (Fig. 3). Gly¹⁸⁸ is located on the loop between $\beta 5$ and $\alpha 5$, which passes below the pyrimidine ring (Fig. 3). The 3'-OH of TtPRODHD is also directed toward this loop but does not form hydrogen bonds with the loop. The 4'-OH is approximately trans to 3'-OH, and it forms a hydrogen bond to the pyrophosphate (3.0 Å, Fig. 4B). In contrast, in PutA86-669 the 2'-OH and 3'-OH groups are rotated $\sim 90^\circ$ from the corresponding groups of TtPRODHD (Fig. 4B). Moreover, the 4'-OH of PutA86-669 is rotated 180° from that of TtPRODHD and forms an intramolecular hydrogen bond with a ribose hydroxyl (Fig. 4B).

As noted above, the adenine bases of TtPRODHD and PutA86-669 are separated by 13 Å. This is due to different dihedral angle rotations of the pyrophosphate (Fig. 4B). In TtPRODHD, the adenine sits atop the N-terminal ends of $\alpha 6$ and $\alpha 7$ (Fig. 1A, the right side provides the best view). In contrast, the adenine of PutA86-669 packs against $\alpha 5a$ and forms an intimate stacking interaction with Trp⁴³⁸ (Fig. 2A).

Three factors contribute to the strikingly different FAD conformations in TtPRODHD and PutA. First, Asp²²⁸ of TtPRODHD replaces Asn⁴⁸⁸ of PutA. Although this seems like a conservative change, the structural ramifications are significant. Asn⁴⁸⁸ forms a hydrogen bond to the FAD pyrophosphate in PutA (Fig. 5A). Changing this residue to Asp not only eliminates the hydrogen bond but also creates electrostatic repulsion between the Asp²²⁸ carboxylate and the pyrophosphate. Apparently, this electrostatic clash is avoided in TtPRODHD by rotation of the pyrophosphate dihedral angle so that the adenosine half of the pyrophosphate faces away from Asp²²⁸ (Fig. 5A). Note that

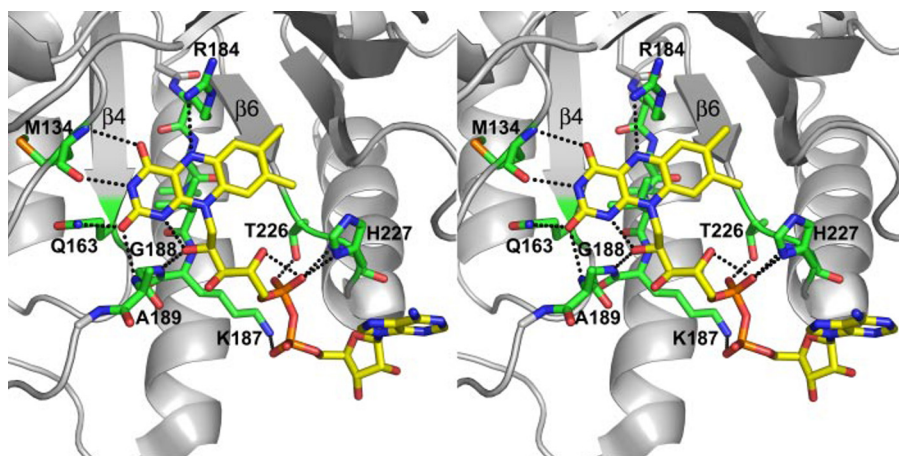


FIGURE 3. Stereographic drawing of protein-FAD interactions in TtPROD H. The dotted lines indicate hydrogen bonds and ion pairs. The FAD is yellow. Residues interacting with FAD are green. Strands $\beta 4$ and $\beta 6$ are indicated.

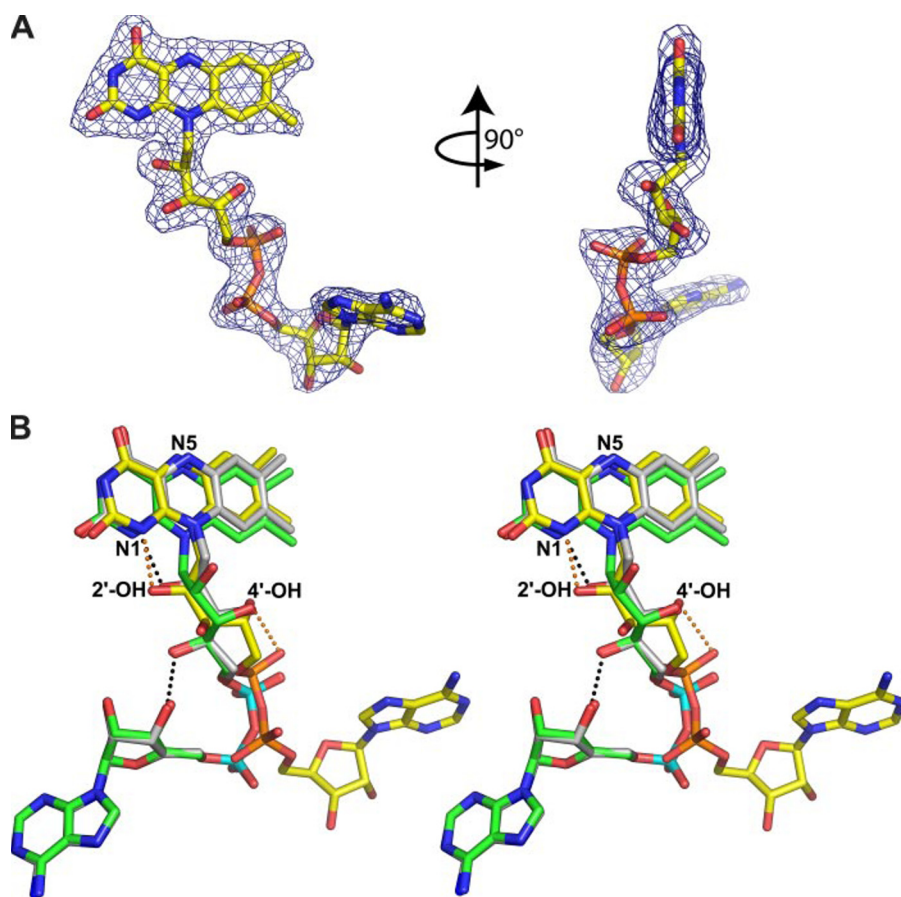


FIGURE 4. Comparison of the FAD conformations of TtPROD H and PutA86-669. A, two views of the FAD from TtPROD H covered by an experimental electron density map (1σ). The map was calculated using F_{obs} and experimental phases after density modification. B, stereographic view of the FAD cofactors from TtPROD H (yellow, C atoms; orange, P atoms), PutA86-669/THFA (green, C atoms; cyan, P atoms) and dithionite-reduced PutA86-669 (white, C atoms; cyan, P atoms). The 2'-OH and 4'-OH groups of TtPROD H are indicated. Note that the 2'-OH groups of TtPROD H and dithionite-reduced PutA86-669 superimpose nearly perfectly, whereas the 2'-OH group of PutA86-669/THFA points toward the viewer. The orange dotted lines denote hydrogen bonds in the TtPROD H cofactor. The black dotted lines denote hydrogen bonds in the PutA86-669 cofactors. The 3'-OH hydrogen bond to the ribose is present in both PutA86-669 structures, but only one dotted line is drawn for clarity.

Asp²²⁸ forms an ion pair with Lys¹⁸⁷, which in turn, forms an ion pair with the pyrophosphate (Fig. 5A). It is concluded that Asp²²⁸ helps set up the FAD conformation in TtPROD H by

electrostatic repulsion with the pyrophosphate and electrostatic attraction to Lys¹⁸⁷.

The other two factors contributing to the different FAD conformations are two helices that are present in PutA but absent in TtPROD H. As noted in the previous section, PutA $\alpha 5a$ is replaced by a short loop in TtPROD H (Fig. 5A). Thus, TtPROD H does not have a residue equivalent to PutA Trp⁴³⁸ for stacking against the adenine (Fig. 5A). In addition, PutA86-669 has extra secondary structural elements following $\alpha 8$, owing to its longer polypeptide length. These extra elements are presumably important for linking the PROD H and P5CDH domains of PutA. One of these elements, a helix formed by PutA residues 565-570, is found to clash with the adenosine of TtPROD H when the two structures are superimposed (clash distances $<1\text{ \AA}$, Fig. 2A). Thus, steric considerations prevent PutA86-669 from accommodating the FAD in the conformation observed in TtPROD H.

The aforementioned analysis compared conformations of oxidized cofactors from TtPROD H and PutA86-669. We recently reported the structure of dithionite-reduced PutA86-669 and showed that reduction of the FAD induces a 22° bend of the isalloxazine and rotation of the 2'-OH by 90° so that it is tucked under the pyrimidine ring and forms a hydrogen bond to the FAD N1 (39). Thus, the conformations of the 2'-OH groups of oxidized TtPROD H and reduced PutA86-669 are identical (Fig. 4B, compare yellow and white FADs). Electron density maps show that the isalloxazine ring in TtPROD H is planar (Fig. 4A, right), as is the case for oxidized PutA86-669 (3, 4), so it is unlikely that TtPROD H was reduced by exposure to X-rays during data collection. We conclude that the TtPROD H structure presented here represents the conformation of the oxidized enzyme.

Thus, in terms of the 2'-OH conformation, the oxidized cofactor of TtPROD H resembles the reduced cofactor of PutA86-669 (Fig. 4B). This result has implications for understanding the

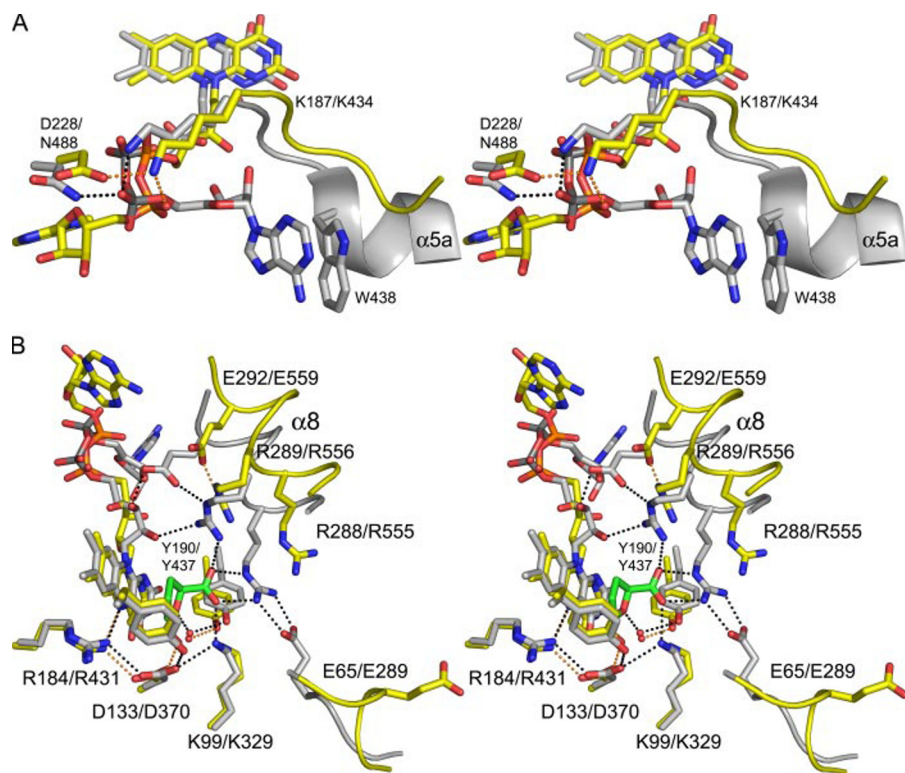


FIGURE 5. Comparison of the active sites of TtPRODH (yellow) and PutA86–669/THFA (white). A, stereographic drawing emphasizing differences between the two enzymes in interactions with the pyrophosphate and adenosine moieties. Residues are labeled as TtPRODH/PutA86–669. The orange and black dotted lines denote hydrogen bonds in TtPRODH and PutA86–669, respectively. B, stereographic drawing showing differences between the two enzymes in the proline binding pocket. The inhibitor THFA is shown in green. Residues are labeled as TtPRODH/PutA86–669. The orange and black dotted lines denote hydrogen bonds in TtPRODH and PutA86–669, respectively. For clarity, Tyr²⁷⁵/Tyr⁵⁴⁰, which form hydrogen bonds with Asp¹³³/Asp³⁷⁰, are not labeled.

differences in membrane association of *E. coli* PutA and TtPRODH (see “Discussion”).

Ligand-free, Solvent-exposed Active Site—It is significant that the TtPRODH active site does not contain a bound proline analog, because all PutA86–669 crystal structures solved to date have active site ligands bound, including acetate (PDB code 1TJ2), L-lactate (PDB code 1TJ0), THFA (PDB code 1TIW), and hyposulfite (PDB code 2FZM). Thus, the TtPRODH structure provides the first view of a ligand-free PRODH active site. Because the amino acid sequences of the proline binding pockets are identical in TtPRODH and *E. coli* PutA, comparison of ligand-free TtPRODH and inhibitor-bound PutA86–669 possibly provides insights into conformational changes induced by the binding of proline. The following analysis refers to molecule A of the asymmetric unit, but it also holds true for molecule B.

There are two major differences between TtPRODH and PutA86–669 in the region of the proline-binding pocket. First, $\alpha 8$ is shifted 3–4 Å away from the proline-binding pocket in TtPRODH (Figs. 2A and 5B). The consequences of this conformational difference are significant, because $\alpha 8$ contributes two absolutely conserved Arg residues that bind the substrate. As shown in Fig. 5B, the two conserved Arg side chains of PutA86–669 (Arg⁵⁵⁵ and Arg⁵⁵⁶) form ion pairs with the carboxyl group of THFA. Movement of $\alpha 8$ in TtPRODH pulls the analogous residues (Arg²⁸⁸ and Arg²⁸⁹) out of the proline-binding pocket

(Fig. 5B). Interestingly, there is a conserved glycine residue in the loop connecting $\beta 8$ and $\alpha 8$ (Fig. 1A, Gly²⁷⁹). It is possible that that Gly²⁷⁹ facilitates movement of $\alpha 8$ by serving as a flexible hinge.

The second major difference between the proline binding pockets of TtPRODH and PutA86–669 involves a conserved ion pair that is present in PutA86–669 but absent in TtPRODH. As shown in Fig. 5B, Arg⁵⁵⁵ of PutA86–669 forms an ion pair with conserved Glu²⁸⁹. The analogous ion pair of TtPRODH (Arg²⁸⁸–Glu⁶⁵) is not observed because Glu⁶⁵ is flipped out of the active site and points into the solvent (Fig. 5B). This difference is quite large, with the carboxyl groups of TtPRODH Glu⁶⁵ and PutA Glu²⁸⁹ separated by 10 Å. We note that Glu⁶⁵ is free of crystal contacts in both molecules, and so the observed conformation of this residue is not likely an artifact of crystal packing.

Because of the conformational differences involving $\alpha 8$ and Glu⁶⁵ (Glu²⁸⁹) the active site of TtPRODH is open and the FAD is solvent-exposed, which sharply contrasts the

closed active site of PutA86–669/THFA in which the FAD is buried. The solvent-accessible surface area (SASA) of the FAD in TtPRODH is 220 Å² for the A chain and 275 Å² for the B chain, based on analysis with the Ligand Protein Contacts server (40). For reference, the SASA of the isolated FAD is 963 Å². By contrast, the FAD in PutA86–669/THFA is buried, with SASA of only 25 Å². The THFA is also buried (0.0 Å² SASA). Exposure of the FAD in TtPRODH can be appreciated by viewing a space-filling model of the enzyme (Fig. 2B). It is evident from this view that almost the entire surface of the isoalloxazine *si* face is solvent-exposed in TtPRODH. In particular, note that the hydride transfer acceptor atom of the flavin, N5, is open to solvent in TtPRODH. This atom is completely buried in PutA86–669/inhibitor complexes.

Absorbance Spectroscopy and Steady-state Kinetics—We performed biochemical analyses of TtPRODH to understand how monofunctional PRODHs differ from PutA PRODH domains in terms of spectroscopic and steady-state kinetic properties. TtPRODH displayed a flavin absorption spectrum similar to that of previously characterized PutAs (41) with maxima at 381 nm and 452 nm (Fig. 6A). Potentiometric titration of TtPRODH yielded an E_m value of –75 mV for the bound FAD cofactor (Fig. 6A), which is similar to that previously reported for *E. coli* PutA (E_m = –77 mV, pH 7.5) (42). No significant stabilization of semiquinone species was observed during the titration (Fig. 6A). Titration of TtPRODH with proline under

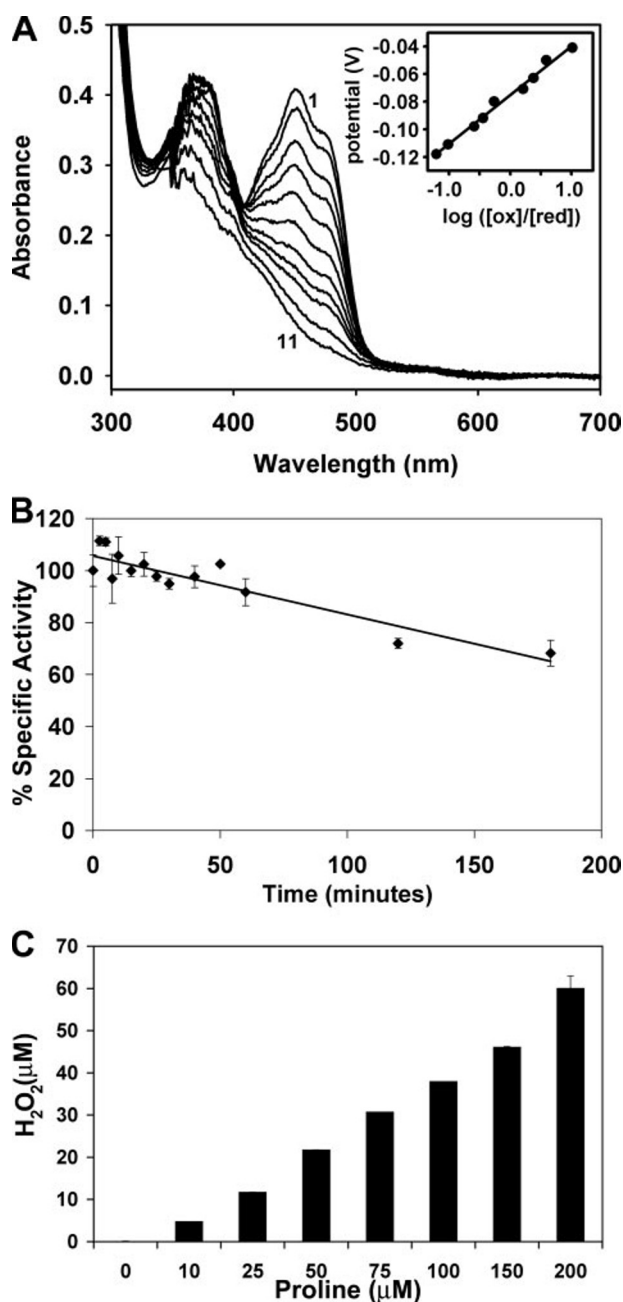


FIGURE 6. **Biochemical studies of TtPRODH.** A, potentiometric titration of TtPRODH (25 μM) at 20 °C (pH 7.5). Curves 1–11 correspond to fully oxidized, –0.041 V, –0.050 V, –0.063 V, –0.071 V, –0.08 V, –0.092 V, –0.098 V, –0.111 V, –0.118 V, and fully reduced, respectively. The inset is a Nernst plot of the potentiometric data, which yielded a midpoint potential of $E_m = -0.075$ V with a slope of 35 mV. B, thermostability analysis of TtPRODH. The percent activity remaining after incubation of the enzyme at 90 °C is plotted as a function of incubation time. C, generation of hydrogen peroxide by TtPRODH. Production of H_2O_2 was measured using the Amplex Red H_2O_2 /peroxidase assay kit as described in the text.

aerobic conditions yielded an apparent equilibrium constant for the formation of TtPRODH_{red}-P5C (Reaction 1) of 4.5 mM^{-1} proline, which is ~2-fold lower than the value of 9.5 mM^{-1} for *E. coli* PutA (34).

The kinetic parameters of TtPRODH using proline as the substrate were estimated to be $K_m = 27$ mM, $V_{\text{max}} = 20.5$ units/mg and $k_{\text{cat}} = 13$ s^{-1} (Table 2). For reference, the corresponding values for PutAs from *E. coli* and *Helicobacter pylori*

TABLE 2

Kinetic parameters using L-proline as the substrate

Assay	TtPRODH	<i>E. coli</i> PutA ^a	<i>H. pylori</i> PutA ^b
Proline:DCPIP assay			
K_m (mM)	27	100	146
V_{max} (units/mg of protein)	20.5	5	3.6
k_{cat} (s^{-1})	13	12	8
k_{cat}/K_m ($\text{s}^{-1} \text{M}^{-1}$)	481	122	56
K_i for THFA (mM)	1.0	0.2	0.35
Proline:O_2 assay			
K_m (mM)	1.3	ND ^c	150
V_{max} (milliunits/mg of protein)	335	<2	230
k_{cat} (min^{-1})	12.7	<0.3	31
DCPIP/ O_2 activity ratio	61	>2500	16

^a Data were obtained from Zhu *et al.* (42) and Krishnan and Becker (44).

^b Data were obtained from Krishnan and Becker (44).

^c ND, not determined.

are listed in Table 2. The K_m parameter of TtPRODH is 3–5 times lower than those of the PutAs, whereas the k_{cat} value is comparable to the PutA values (Table 2). The kinetic parameters for TtPRODH are closer to those of PutA86–669 ($K_m = 60$ mM, $k_{\text{cat}} = 17$ s^{-1}) than those of the PutAs.

Interestingly, 3,4-dehydro-L-proline was more efficiently oxidized than proline, with $K_m = 4$ mM, $V_{\text{max}} = 119$ units/mg, and $k_{\text{cat}} = 75$ s^{-1} . Analogous data for PutA are not available, although Wood reported that in *E. coli* K12 PutA detoxifies 3,4-dehydro-L-proline by oxidation (43).

THFA, L-mandelate, and L-lactate were identified as competitive inhibitors of TtPRODH with K_i values of 1.0, 2.4, and 2.5 mM, respectively. THFA is also a competitive inhibitor of PutA86–669, *E. coli* PutA, and *H. pylori* PutA ($K_i = 0.2$ –0.3 mM). Other proline analogs such as trans-4-hydroxy-L-proline, cis-4-hydroxy-L-proline, L-azetidine-2-carboxylic acid, and L-pipecolic acid were neither inhibitors nor substrates of TtPRODH.

TtPRODH is highly thermostable, based on residual activity measurements, as expected for an enzyme from an extreme thermophile. TtPRODH exhibited over 85% residual activity after a 1-h incubation at 90 °C, and over 60% residual activity after 3 h (Fig. 6B). The half-life estimated from these data was 257 min. For comparison, PutA86–669 exhibits a 50% drop in activity after 2 h at 45 °C (3).

Proline: O_2 Reactivity and Generation of ROS—The solvent-exposed active site of TtPRODH suggested the possibility that the reduced enzyme could react with molecular oxygen. A chromogenic assay based on *o*-AB was used to measure the rate of oxygen reactivity. In this assay, P5C produced by oxidation of proline forms a complex with *o*-AB, which is monitored by absorbance at 443 nm. Detection of the *o*-AB-P5C complex over time indicates reactivity of the reduced FAD with O_2 during catalytic turnover with proline.

TtPRODH exhibited significant proline: O_2 reactivity with kinetic parameters of $K_m = 1.3$ mM, $V_{\text{max}} = 335$ milliunits/mg, and $k_{\text{cat}} = 12.7$ min^{-1} (Table 2). Data from the *o*-AB assay are typically expressed as the ratio of the specific activity from the DCPIP assay to the specific activity from the *o*-AB assay, which indicates the preference of utilizing DCPIP over O_2 as the electron acceptor. For TtPRODH, this ratio was 61. For comparison, the corresponding ratios for PutAs from *H. pylori* and *E. coli* were 16 and >2500, respectively (Table 2).

TABLE 3

Conserved sequence motifs of PutAs and bacterial monofunctional PRODHs

Letters in bold indicate amino acid residues that are identically conserved throughout the entire family of bacterial PutAs and PRODHs. Uppercase letters denote residues that are identically conserved within a branch of the family. Lowercase letters indicate amino acid residues that are highly, but not identically, conserved within a branch.

Motif	Branch 1 ^a	Branch 2 ^b	Branch 3 ^c
1	s-y-D-m-L-G-E (288, 289)	v-d-l-l-G-E (146, 147)	l-d-X-l-G-E (64, 65)
2	i-S-i/v-K-l-S (329)	s-i/v-K-X-t/s (177)	S-i/v/l-K-X-s/t (99)
3	i-D-A-E-E (370, 372)	D-M-E (224, 226)	D-m-E (133, 135)
4	G-X-v-v-Q-a-y-q-k-R (404)	g-i-v-Q-A-Y-l-X-d (254)	G-X-v-l-Q-a-y-L (163)
5	R-L-v-K-G-A-Y-W-D-X-E-i-k (431, 434, 435)	R-l-V-K-G-a-y-w-d (285, 288, 289)	R-l-v-K-G-a-Y-X-E-p (184, 187, 188)
6	F-a-t-H-N (487)	i-a-s-H-N (342)	i-a-t-H-D (227)
7	E-f-Q-r-l-h-G-M-g-e (513, 515)	e-f-q-v-L-y-G-M-a (371, 373)	E-f-Q-m-L-y-G-i-r (254, 256)
8	i-Y-a-P-v-G (540)	l-Y-x-y-P-g (391)	v-Y-v-p-y-G (275)
9	L-L-a-Y-L-v-R-R-l-l-E-N-G (552, 555, 556, 559)	Y-L-v-R-R-l-l-E-N (406, 409, 410, 413)	Y-f-m-R-R-l-a-E (285, 288, 289, 292)

^a Numbers in parenthesis refer to *E. coli* PutA residue numbers.

^b Numbers in parenthesis refer to *H. pylori* PutA residue numbers.

^c Numbers in parenthesis refer to TtPRODH residue numbers.

Generation of ROS by TtPRODH was also studied. The production of superoxide was examined by monitoring reduction of cytochrome *c* during the proline oxidation catalytic cycle. Reduction of cytochrome *c* was observed, which is consistent with generation of superoxide. Addition of superoxide dismutase to the assay (30 μ g) eliminated the observed effect, which further implicates superoxide as the product of the proline:O₂ reaction. Because superoxide decomposes in water to H₂O₂, the generation of H₂O₂ was also examined. H₂O₂ was produced in a proline-dependent manner (Fig. 6C).

Conserved Sequence-Structure Motifs of the PRODH Family—PRODH and P5CDH are separate enzymes in some organisms, whereas the two are fused in other organisms. The traditional view, which was developed prior to the whole genome sequence era, has been that PRODH and P5CDH appear as separate enzymes in eukaryotes and as fused bifunctional enzymes (PutA) in bacteria. Our analysis of genome sequence data, however, reveals a more complex situation for bacteria. The updated view is that PutAs are indeed restricted to bacteria, but monofunctional PRODHs and P5CDHs appear in both eukaryotes and bacteria.

The distribution of PutAs and monofunctional PRODHs in bacteria is depicted in the phylogenetic tree in supplemental Fig. S1. Three main branches are evident. Branch 1 contains the best characterized PutAs, including PutAs from *E. coli*, *Bradyrhizobium japonicum*, and *Salmonella typhimurium*. The organisms represented in branch 1 are primarily α -, β -, and γ -proteobacteria. The PutAs in branch 1 of our data set have 999–1361 residues, and the pairwise sequence identities are 38–99% with an average of 49%.

Branch 2 contains PutAs from Gram-negative cyanobacteria, δ - and ϵ -proteobacteria, and corynebacterium. The polypeptide length for branch 2 PutAs is 982–1294, and the pairwise sequence identity range is 23–73% with an average identity of 38%. Note that branch 2 is divided into two distinct groups, denoted 2A and 2B in supplemental Fig. S1. PutAs from branch 2 have only recently been explored. For example, Krishnan and Becker (44) showed that branch 2A PutAs from *H. pylori* and *Helicobacter hepaticus* are unique among PutAs in that they exhibit oxygen reactivity and generate proline-dependent ROS.

Branch 3 consists entirely of monofunctional PRODHs. TtPRODH is the only enzyme of this branch to be purified and characterized. PRODHs of branch 3 have 279–333 amino acid residues, and the pairwise sequence identities for this group are

23–79% with an average of 38%. In some branch 3 organisms, the PRODH and P5CDH genes are very close together. For example, in *T. thermophilus*, only 15 bases separate the stop codon of the PRODH gene from the Met start codon of the P5CDH gene. In other organisms the two genes are quite far apart. For example, in *Staphylococcus aureus* subsp. *aureus* Mu50, the PRODH and P5CDH genes are separated by 800 kb. Interestingly, most of the organisms represented in branch 3 are Gram-positive bacteria. Counterexamples include *Thermus*, *Solibacter*, *Salinibacter*, and *Chlorobium*.

The availability of two PRODH structures (TtPRODH and PutA86–669) and many sequenced bacterial genomes allowed analysis of conserved sequence-structure motifs of the bacterial PutA/PRODH family. Nine conserved motifs were identified by mapping multiple sequence alignments onto the TtPRODH and PutA86–669 structures (Table 3). Each motif contains at least one residue that is identically conserved throughout the entire bacterial PutA/PRODH family (Table 3, bold letters).

The nine motifs cluster near the active site, and the identically conserved residues of the motifs have important roles in FAD binding and substrate recognition. Motifs 4–6 are primarily involved in FAD binding. For example, the conserved Gln of motif 4 forms a hydrogen bond to the FAD O2 (Fig. 3, Gln¹⁶³), whereas the Arg of motif 5 forms a hydrogen bond to the FAD N5 (Fig. 3, Arg¹⁸⁴). Also, the Lys and His of motifs 5 and 6 interact with the pyrophosphate (Fig. 3, Lys¹⁸⁷ and His²²⁷).

Motifs 1–3 and 7–9 are responsible for substrate recognition. For example, the Lys and Arg side chains of motifs 2 and 9 form ion pairs to the substrate carboxyl, as shown for the PutA86–669/THFA structure (Fig. 5B, PutA residues Lys³²⁹, Arg⁵⁵⁵, and Arg⁵⁵⁶). The conserved Glu residues of motifs 1 and 9 play indirect roles in substrate recognition by providing stabilizing ion pairs to the two Arg residues of motif 9 when the substrate/product is bound (Arg⁵⁵⁵–Glu²⁸⁹ and Arg⁵⁵⁶–Glu⁵⁵⁹ in PutA, Fig. 5B). Size and shape complementarity are enforced by non-polar contacts between the substrate and the Leu and Tyr side chains of motifs 7–9 (3). Finally, the conserved Asp of motif 3 (Asp¹³³ of TtPRODH) plays a dual role in substrate recognition and FAD binding by forming stabilizing interactions with the Lys of motif 2 (substrate recognition, Fig. 5B), Tyr of motif 8 (substrate recognition, Fig. 5B) and the conserved Arg of motif 5 (FAD binding, Fig. 5B).

DISCUSSION

New Subfamily of PRODH—The genesis of this work was the realization, based on analysis of genome sequence data, that some bacteria lack *putA* genes and instead encode PRODH and P5CDH as separate monofunctional enzymes. This observation is significant because the traditional view of proline catabolism was that monofunctional enzymes are restricted to eukaryotes. Thus, bacterial monofunctional PRODHs represent a new subfamily of proline catabolic enzyme. Moreover, isolation of recombinant eukaryotic PRODHs in sufficient quantity and purity for biophysical study has been problematic,³ and therefore the bacterial homologs are potentially attractive model systems for understanding human PRODH. We thus set out to characterize a bacterial monofunctional PRODH to establish paradigms for this new subfamily, compare its structure and biochemical properties to those of PutAs, and set the stage for probing interactions between PRODH and P5CDH.

One major result of this work is that monofunctional PRODHs and PutAs share a common catalytic core consisting of a unique TIM barrel (Fig. 2A). The PutA/PRODH barrel is distinguished from other TIM barrels by placement of $\alpha 8$ above the barrel. This distortion is functionally significant because $\alpha 8$ contributes conserved motif 9 to the active site (Table 3). Observation of this structural distortion in enzymes from two branches of the bacterial PutA/PRODH family suggests that it is a defining structural signature of this family.

Interestingly, the TIM barrel catalytic core, FAD conformations, and proline binding pocket described here for the PutA/PRODH family bear no resemblance to those of PRODHs from the hyperthermophilic archaeon *Pyrococcus horikoshii* (PDH1 and PDH2). For example, PDH1 is an ($\alpha\beta$)₄ hetero-octameric complex with the β subunit binding an FAD cofactor and exhibiting PRODH activity (6). The crystal structure of PDH1 shows that the FAD-binding domain has a Rossmann dinucleotide-binding fold similar to that of monomeric sarcosine oxidase, which is a member of the glutathione reductase family (6). As expected for a Rossmann fold protein, the FAD of PDH1 is highly extended, and the pyrophosphate interacts with a glycine-rich loop and associated conserved water molecule (45). Thus, the protein-FAD interactions in PDH1 are quite different from those described here. Furthermore, proline binds on the *re* face of the FAD in PDH1, in contrast to PutA and TtPRODH, which bind proline on the *si* face. Thus, the PutA/PRODH and PDH1 families represent two distinctly different solutions to the problem of catalyzing the oxidation of proline by a flavoenzyme.

FAD Conformation—Although the catalytic cores of TtPRODH and PutA86–669 are similar in overall fold, the FAD conformations are surprisingly different. This difference is attributed to an Asp/Asn sequence difference (Fig. 5A) and two helices present in PutA86–669 but absent in TtPRODH (Figs. 2A and 5A). Based on amino acid sequence alignments, the Asn-pyrophosphate hydrogen bond, $\alpha 5a$, and the 560s helix are present in all PutAs and missing in all bacterial monofunctional PRODHs. Interestingly, the stacking Trp of $\alpha 5a$ is pres-

ent in most, but not all, PutAs. For example, branch 2A enzymes have Met or Leu in place of the stacking Trp, but these residues could also provide a non-polar interaction with the FAD adenine. We therefore predict that all PutAs have the FAD conformation observed in PutA86–669 and all bacterial monofunctional PRODHs have the FAD conformation observed in TtPRODH. It is concluded that the FAD conformation is the major structural difference between PutAs and bacterial monofunctional PRODHs.

The different FAD conformations in PutA and monofunctional PRODHs presumably reflect different structural and functional requirements. For example, we suggest that the difference in position of the ribityl 2'-OH reflects the different membrane association requirements of *E. coli* PutA and TtPRODH. *E. coli* PutA is membrane-associated only when the FAD is in the reduced state. In the oxidized state, it remains in the cytoplasm and represses transcription of the *putA* and *putP* (encodes a proline transporter) genes (15, 46, 47). We recently showed that reduction of FAD in PutA86–669 triggers rupture of the hydrogen bond between the ribityl 2'-OH and Arg⁵⁵⁶, causing rotation of the 2'-OH so that it is tucked below, and hydrogen bonded to, the FAD N1 (39). We further showed that the 2'-OH-Arg⁵⁵⁶ hydrogen bond is a structural constraint that prevents oxidized PutA from binding the membrane (39). Interestingly, we find here that the 2'-OH of oxidized TtPRODH is tucked below the FAD N1, that is, cocked in the membrane-binding position (Fig. 4). This makes sense because TtPRODH does not have a repressor function and is presumably membrane-associated in both the oxidized and reduced states.

A larger question is why the global FAD conformations of PutAs and monofunctional PRODHs differ so much, with the adenosine moieties of the two cofactors separated by 13 Å. It is possible that the particular FAD conformation found in PutA is necessary for coordinating the two catalytic functions of PutA, *i.e.* substrate channeling. Monofunctional PRODH and P5CDH may interact and exhibit intermolecular substrate channeling according to the Rosetta Stone hypothesis. We note that this occurs for tryptophan synthase in which indole is channeled between separate α and β subunits (16, 17). Presumably the FAD conformation in TtPRODH is the one required for docking of TtPRODH with TtP5CDH. Observation of different FAD conformations in TtPRODH and PutA86–669 may indicate that the protein-protein interface between TtPRODH and TtP5CDH differs substantially from the PRODH:P5CDH domain interface in PutA. Clearly, structures of full-length PutAs, other monofunctional PRODHs, and a monofunctional PRODH:P5CDH complex would address the issue of why PutAs and monofunctional PRODHs require such different FAD conformations.

Relationships between Bacterial and Human PRODH—Humans have two isozymes of PRODH, which share about 50% amino acid sequence identity (48). PRODH1 is encoded on chromosome 19, is expressed almost exclusively in liver and kidney, and catalyzes oxidation of L-hydroxyproline. PRODH2 is encoded on chromosome 22q11, is expressed more widely than PRODH1 (brain, heart, pancreas, kidney, and liver), and specifically oxidizes L-proline.

³ T. A. White, N. Krishnan, D. F. Becker, and J. J. Tanner, unpublished results.

PRODH2 is part of the p53 signaling pathway with the PRODH2 gene identified as a p53-inducible gene (49). Up-regulation of PRODH2 and proline oxidation in lung, renal, and colon carcinoma cells has been shown to generate ROS and induce cell death by mitochondria-dependent processes (50–55). Perturbation of mitochondrial membranes by ROS causes the release of cytochrome *c* into the cytosol and subsequent activation of the intrinsic caspase pathway. Because the p53-apoptosis pathway involves ROS, PRODH2 appears to play a role in p53-mediated apoptosis by modulating the cellular redox environment. Indeed, antisense repression of PRODH2 prevents p53-induced apoptosis (55). Thus, PRODH2 is a pro-apoptotic protein that helps reduce carcinogenesis in humans by serving as a ROS generator. Accordingly, PRODH2 is often referred to as proline oxidase in the literature.

Bacterial monofunctional PRODHs are potentially attractive model systems for understanding the structure and biochemical function of human PRODHs. Although the sequence identity between TtPRODH and human PRODH1/2 is <20%, conserved motifs 3–9 are clearly present in the sequences of both human PRODH1 and PRODH2. Considering residues within 10 Å of the active site, there is 45% sequence conservation between human PRODH1/2 and TtPRODH. These data strongly suggest that the human enzymes have the ($\beta\alpha$)₈ catalytic core and active site structure common to TtPRODH and PutA86–669. On the other hand, human PRODH1 and PRODH2 have 536 and 600 residues, respectively, compared with only 307 for TtPRODH, and so the human enzymes clearly have additional structural elements not found in TtPRODH.

Interestingly, analysis of human PRODH1/2 amino acid sequences suggests that the conformation of FAD in human PRODH1/2 is similar to that of PutA rather than TtPRODH. Both human enzymes have Asn at the position equivalent to *E. coli* PutA Asn⁴⁸⁸. Furthermore, sequence alignments suggest that the human enzymes have the equivalent of PutA α 5a and that Leu replaces the stacking Trp. We note that Leu is also present at this position in some branch 2A PutAs. Thus, the FAD conformation observed in *E. coli* PutA is probably present in the human enzymes, and the conformation observed in TtPRODH seems to be a unique signature of bacterial monofunctional PRODHs.

Production of proline-dependent superoxide is central to the role of PRODH2 in p53-mediated apoptosis (54, 56). It is therefore highly significant that we observed reactivity of TtPRODH with molecular oxygen resulting in superoxide generation.

Reactivity of a FAD-dependent dehydrogenase with molecular oxygen implies that the reduced FAD is accessible to solvent. *E. coli* PutA is essentially non-reactive with molecular oxygen (44), suggesting that the FAD remains sequestered even after the active site opens to release P5C. On the other hand, *H. pylori* PutA reacts strongly with molecular oxygen indicating that O₂ has access to the reduced cofactor (44). Krishnan and Becker (44) have suggested that Asn²⁹¹ of *H. pylori* PutA, which is replaced by the bulkier Tyr⁴³⁷ in *E. coli* PutA, may account for the differences in O₂ reactivity of these two PutAs. This particular Tyr is part of conserved motif 5 (Table 3, Tyr⁴³⁷ in *E. coli* PutA). It forms a water-mediated hydrogen bond to the O atom of the THFA ring (Fig. 5B) and is in position to protect the

active site from bulk solvent by virtue of its location on the edge of the proline-binding site (see Tyr¹⁹⁰ in Fig. 2B). Tyrosine, being larger than Asn, may afford more protection of the active site from solvent. Interestingly, human PRODH2 and TtPRODH, both ROS generators, have Tyr at this position (Tyr⁴⁴⁶ in PRODH2 and Tyr¹⁹⁰ in TtPRODH, Fig. 2B). Thus, the Asn hypothesis, which may explain the difference in O₂ reactivity between *E. coli* and *H. pylori* PutAs, does not explain the O₂ reactivity of TtPRODH and human PRODH2. The TtPRODH structure shows that α 8 is important for protecting the FAD from solvent and that there is sufficient flexibility in the β 8– α 8 loop to allow movement of α 8 away from the proline binding pocket resulting in exposure of the isoalloxazine. These results suggest that movement of α 8 may contribute to O₂ reactivity of PRODHs.

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REFERENCES

- Adams, E., and Frank, L. (1980) *Annu. Rev. Biochem.* **49**, 1005–1061
- Phang, J. M. (1985) *Curr. Top. Cell Reg.* **25**, 92–132
- Zhang, M., White, T. A., Schuermann, J. P., Baban, B. A., Becker, D. F., and Tanner, J. J. (2004) *Biochemistry* **43**, 12539–12548
- Lee, Y. H., Nadaraja, S., Gu, D., Becker, D. F., and Tanner, J. J. (2003) *Nat. Struct. Biol.* **10**, 109–114
- Inagaki, E., Ohshima, N., Takahashi, H., Kuroishi, C., Yokoyama, S., and Tahirov, T. H. (2006) *J. Mol. Biol.* **362**, 490–501
- Tsuge, H., Kawakami, R., Sakuraba, H., Ago, H., Miyano, M., Aki, K., Katunuma, N., and Ohshima, T. (2005) *J. Biol. Chem.* **280**, 31045–31049
- Menzel, R., and Roth, J. (1981) *J. Biol. Chem.* **256**, 9762–9766
- Brown, E., and Wood, J. M. (1992) *J. Biol. Chem.* **267**, 13086–13092
- Muro-Pastor, A. M., and Maloy, S. (1995) *J. Biol. Chem.* **270**, 9819–9827
- Muro-Pastor, A. M., Ostrovsky, P., and Maloy, S. (1997) *J. Bacteriol.* **179**, 2788–2791
- Becker, D. F., and Thomas, E. A. (2001) *Biochemistry* **40**, 4714–4722
- Zhu, W., and Becker, D. F. (2003) *Biochemistry* **42**, 5469–5477
- Zhang, W., Zhou, Y., and Becker, D. F. (2004) *Biochemistry* **43**, 13165–13174
- Krishnan, N., and Becker, D. F. (2005) *Biochemistry* **44**, 9130–9139
- Larson, J. D., Jenkins, J. L., Schuermann, J. P., Zhou, Y., Becker, D. F., and Tanner, J. J. (2006) *Protein Sci.* **15**, 1–12
- Miles, E. W., Rhee, S., and Davies, D. R. (1999) *J. Biol. Chem.* **274**, 12193–12196
- Huang, X., Holden, H. M., and Raushel, F. M. (2001) *Annu. Rev. Biochem.* **70**, 149–180
- Surber, M. W., and Maloy, S. (1998) *Arch. Biochem. Biophys.* **354**, 281–287
- Marcotte, E. M., Pellegrini, M., Ng, H. L., Rice, D. W., Yeates, T. O., and Eisenberg, D. (1999) *Science* **285**, 751–753
- White, T. A., and Tanner, J. J. (2005) *Acta Crystallogr. Sect. F Struct. Biol. Cryst. Commun.* **61**, 737–739
- Doublie, S. (1997) *Methods Enzymol.* **276**, 523–530
- Matthews, B. W. (1968) *J. Mol. Biol.* **33**, 491–497
- Pflugrath, J. W. (1999) *Acta Crystallogr. Sect. D Biol. Crystallogr.* **55**, 1718–1725
- Terwilliger, T. C. (2003) *Methods Enzymol.* **374**, 22–37
- Cowan, K., and Main, P. (1998) *Acta Crystallogr.* **54**, 487–493
- Morris, R. J., Perrakis, A., and Lamzin, V. S. (2002) *Acta Crystallogr. Sect. D Biol. Crystallogr.* **58**, 968–975
- Winn, M. D., Murshudov, G. N., and Papiz, M. Z. (2003) *Methods Enzymol.* **374**, 300–321

28. van Aalten, D. M., Bywater, R., Findlay, J. B., Hendlich, M., Hooft, R. W., and Vriend, G. (1996) *J. Comput. Aided Mol. Des.* **10**, 255–262
29. Potterton, E., Briggs, P., Turkenburg, M., and Dodson, E. (2003) *Acta Crystallogr. Sect. D Biol. Crystallogr.* **59**, 1131–1137
30. Collaborative Computational Project, Number 4 (1994) *Acta Crystallogr. Sect. D Biol. Crystallogr.* **50**, 760–763
31. Berman, H. M., Westbrook, J., Feng, Z., Gilliland, G., Bhat, T. N., Weissig, H., Shindyalov, I. N., and Bourne, P. E. (2000) *Nucleic Acids Res.* **28**, 235–242
32. Dixon, M. (1953) *Biochem. J.* **55**, 170–171
33. Stankovich, M. T. (1980) *Anal. Biochem.* **109**, 295–308
34. Brown, E. D., and Wood, J. M. (1993) *J. Biol. Chem.* **268**, 8972–8979
35. Mezl, V. A., and Knox, W. E. (1976) *Anal. Biochem.* **74**, 430–440
36. Tarpey, M. M., and Fridovich, I. (2001) *Circ. Res.* **89**, 224–236
37. Thompson, J. D., Higgins, D. G., and Gibson, T. J. (1994) *Nucleic Acids Res.* **22**, 4673–4680
38. Walden, H., Bell, G. S., Russell, R. J., Siebers, B., Hensel, R., and Taylor, G. L. (2001) *J. Mol. Biol.* **306**, 745–757
39. Zhang, W., Zhang, M., Zhu, W., Zhou, Y., Wanduragala, S., Rewinkel, D., Tanner, J. J., and Becker, D. F. (2007) *Biochemistry* **46**, 483–491
40. Sobolev, V., Sorokine, A., Prilusky, J., Abola, E. E., and Edelman, M. (1999) *Bioinformatics* **15**, 327–332
41. Vinod, M. P., Bellur, P., and Becker, D. F. (2002) *Biochemistry* **41**, 6525–6532
42. Zhu, W., Gincher, Y., Docherty, P., Spilling, C. D., and Becker, D. F. (2002) *Arch. Biochem. Biophys.* **408**, 131–136
43. Wood, J. M. (1981) *J. Bacteriol.* **146**, 895–901
44. Krishnan, N., and Becker, D. F. (2006) *J. Bacteriol.* **188**, 1227–1235
45. Bottoms, C. A., Smith, P. E., and Tanner, J. J. (2002) *Protein Sci.* **11**, 2125–2137
46. Chen, C. C., and Wilson, T. H. (1986) *J. Biol. Chem.* **261**, 2599–2604
47. Gu, D., Zhou, Y., Kallhoff, V., Baban, B., Tanner, J. J., and Becker, D. F. (2004) *J. Biol. Chem.* **279**, 31171–31176
48. Phang, J. M., Hu, C. A., and Valle, D. (2001) in *Metabolic and Molecular Basis of Inherited Disease* (Scriver, C. R., Beaudet, A. L., Sly, W. S., and Valle, D., eds) pp. 1821–1838, McGraw-Hill, New York
49. Polyak, K., Xia, Y., Zweier, J. L., Kinzler, K. W., and Vogelstein, B. (1997) *Nature* **389**, 300–305
50. Hu, C. A., Donald, S. P., Yu, J., Lin, W. W., Liu, Z., Steel, G., Obie, C., Valle, D., and Phang, J. M. (2006) *Mol. Cell. Biochem.* **295**, 85–92
51. Rivera, A., and Maxwell, S. A. (2005) *J. Biol. Chem.* **280**, 29346–29354
52. Pandhare, J., Cooper, S. K., and Phang, J. M. (2006) *J. Biol. Chem.* **281**, 2044–2052
53. Maxwell, S. A., and Davis, G. E. (2000) *Proc. Natl. Acad. Sci. U. S. A.* **97**, 13009–13014
54. Liu, Y., Borchert, G. L., Donald, S. P., Surazynski, A., Hu, C. A., Weydert, C. J., Oberley, L. W., and Phang, J. M. (2005) *Carcinogenesis* **26**, 1335–1342
55. Maxwell, S. A., and Rivera, A. (2003) *J. Biol. Chem.* **278**, 9784–9789
56. Donald, S. P., Sun, X. Y., Hu, C. A., Yu, J., Mei, J. M., Valle, D., and Phang, J. M. (2001) *Cancer Res.* **61**, 1810–1815
57. Engh, R. A., and Huber, R. (1991) *Acta Crystallogr. Sect. A* **47**, 392–400
58. Laskowski, R. A., MacArthur, M. W., Moss, D. S., and Thornton, J. M. (1993) *J. Appl. Crystallogr.* **26**, 283–291